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Progressive seawater acidification on the Great Barrier Reef continental shelf

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Coral reefs are highly sensitive to ocean acidification due to rising atmospheric CO₂ concentrations. We present 10 years of data (2009–2019) on the long-term trends and sources of variation in the carbon chemistry from two fixed stations in the Australian Great Barrier Reef. Data from the subtropical mid-shelf GBRWIS comprised 3-h instrument records, and those from the tropical coastal NRSYON were monthly seawater samples. Both stations recorded significant variation in seawater CO₂ fugacity (fCO₂), attributable to seasonal, daytime, temperature and salinity fluctuations. Superimposed over this variation, fCO₂ progressively increased by $> 2.0 \pm 0.3 \mu\text{atm year}^{-1}$ at both stations. Seawater temperature and salinity also increased throughout the decade, whereas seawater pH and the saturation state of aragonite declined. The decadal upward fCO₂ trend remained significant in temperature- and salinity-normalised data. Indeed, annual fCO₂ minima are now higher than estimated fCO₂ maxima in the early 1960s, with mean fCO₂ now ~28% higher than 60 years ago. Our data indicate that carbonate dissolution from the seafloor is currently unable to buffer the Great Barrier Reef against ocean acidification. This is of great concern for the thousands of coral reefs and other diverse marine ecosystems located in this vast continental shelf system.

Atmospheric carbon dioxide (CO₂) concentrations are steadily increasing due to human activities, in the present decade at about 2.5 ppm per year¹. Over a quarter of the rising atmospheric CO₂ is being taken up by the oceans. This lowers the pH and changes the carbon chemistry in surface seawaters, a process called ocean acidification². Due to human CO₂ emissions, surface seawater pH is now lower than it has been for more than 800,000 years³, and the associated chemical changes are considered to be irreversible on centennial to millennial time scales^{4,5}. Many studies have shown that ocean acidification, both in isolation and in combination with global warming, causes profound physiological and ecological changes in marine ecosystems, with far more losers than winners^{6–8}. Calcifying marine organisms such as corals and coralline algae are particularly affected, especially during their early life stages, whereas some photosynthetic organisms benefit from the availability of additional inorganic carbon^{9,10}.

Rates of changes in seawater carbon chemistry vary substantially across regions, and depend not only on atmospheric CO₂ concentrations, but also on local physical and biological factors^{2,11–13}. In coastal, shelf and marginal seas the variation in seawater carbon chemistry is typically much higher than in the open oceans due to regional metabolic processes (photosynthesis/respiration and calcification). These are sometimes accentuated by terrestrial inputs of carbon, alkalinity, nutrients, freshwater and sediments that directly alter the seawater carbon chemistry and also stimulate biological productivity¹⁴. In a process called ‘coastal acidification’, acidification may accelerate in areas of eutrophication^{2,15}. Coastal CO₂ concentrations are also affected by the upwelling of CO₂ rich deeper waters, thermocline shallowing, stratification, freshening and El Niño–Southern Oscillation (ENSO) dynamics. On the other hand, the dissolution of carbonate seafloor sediments can also dampen coastal acidification, depending on carbonate types, grain size and physical condition^{16–18}.

High-precision time series data of changes in the carbon chemistry of seawater are available from an increasing number of oceanographic monitoring stations, often above deep waters (> 500 m), but also increasingly so from coastal waters^{12,13,19,20}. Many of these data series show increases in seasonally corrected surface CO₂, at rates similar to that of CO₂ in the atmosphere¹⁹. Some stations also show reductions in seawater saturation states of aragonite and calcite, albeit with greater regional differences. In most instances large seasonal variation or decadal oceanographic features require longer observation periods before the detection of significant trends are expected¹². For example, diel CO₂ variation of 30–200 $\mu\text{atm day}^{-1}$ and at times over 700 $\mu\text{atm CO}_2$ have been

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recorded for some shallow tropical coastal and shelf marine habitats^{21,22}, a variation that is several-fold greater than in the open waters. Given the large temporal and spatial variability and the high ecological and economic value, more long-term carbon chemistry data from coastal and shelf locations are urgently needed.

In this study, we assessed variation and trends in CO₂ and other proxies for ocean acidification from two sites in the Australian Great Barrier Reef (GBR). The GBR is a shallow carbonate rich continental shelf system extending over 2300 km along the north-eastern Australian coast from 12 to 24° S latitude, and varies in width from 40 to > 200 km (Fig. 1). The GBR is an interesting case study for ocean acidification research, due to its complex hydrodynamic and geomorphological features, and its valuable reef ecosystems. About 5% of the GBR surface is occupied by ~ 3000 coral reefs that are built by corals and other calcifying organisms with a strong dependence on the saturation state of carbonate minerals^{8,10,22}. Furthermore much of the seafloor between the coral reefs is covered by biogenic carbonate sediments²³ with habitats that include mesophotic coral shoals, *Halimeda* mounds, seagrass meadows, sponge gardens and soft bottom ecosystems. Spatial and temporal patterns in the carbon chemistry of the GBR seawater are characterised by substantial cross-shelf gradients, with additional regional features along the coast^{24–26} and in the proximity of coral reefs^{22,27,28}. The surface waters flowing from the Coral Sea onto the GBR shelf are low in nutrient and chlorophyll concentrations, but occasional upwelling onto the shelf break can inject nutrients, total alkalinity and dissolved inorganic carbon into the GBR waters^{29,30}. Thirty-five major rivers drain into the GBR lagoon, and their sediment and nutrient loads affect water clarity for months after floods^{31,32}. Terrigenous sediments dominate near the mouths of these rivers^{33,34}. The ratio of terrigenous to carbonate sediments declines steeply away from the coast, as flood plumes disperse predominantly the fine silts, and storms redistribute terrigenous sediments predominantly within the innermost 15 km of the lagoon³⁵.

We present 10 years (2009–2019) of carbon chemistry data together with auxiliary data, from two fixed long-term GBR oceanographic monitoring stations ~ 650 km apart (Fig. 1a, Supplementary Table S1). The first station, GBRWIS^{36,37} (– 23.459° S, 151.927° E), is located in the subtropical southern mid-shelf GBR, at ~ 16 m depth in a channel separating the coral reefs of Wistari Reef from those surrounding Heron Island and ~ 20 km from the edge of the continental shelf (Fig. 1b). The second station, NRSYON³⁸ (– 19.305° S, 147.622° E), is a National Reference Station in the tropical central coastal GBR, located at ~ 26 m depth at the Yongala shipwreck near the Burdekin River mouth. We determined the long-term trend and variation in CO₂, and identified the main physical and chemical drivers of the observed changes. We assessed commonalities and differences of the two stations, and we hindcast changes over the last 60 years. Despite their contrasting settings and substantial seasonal and diel variation, both GBR stations showed significant and rapid rate of increase in CO₂, hence it is likely that some biological processes in GBR coral reefs are already affected by ocean acidification.

Results

GBRWIS station. At GBRWIS, all four instrumental time series showed strong long-term trends, which we refer to in the rest of this paper as the decadal trend. Atmospheric CO₂ increased at a mean rate of 2.00 ppm year^{–1} ± 0.011 (1 standard error) throughout the 10-year observation period, with only minor seasonal variation (Fig. 2a, Table 1). Mean daily seawater CO₂ fugacity (*f*CO₂) increased progressively by 6.4% of its initial value over the decade (Fig. 2b), at a mean rate of 2.35 ± 0.17 μatm year^{–1}. The rates of increase in *f*CO₂ did not differ significantly from that of the atmospheric CO₂ (GLS on monthly averaged data, N = 109, P = 0.69). Mean daily seawater temperature increased slightly throughout the decade by 0.026 ± 0.013 °C year^{–1}, and there was also an upward trend in salinity at a mean rate of 0.037 ± 0.001 units year^{–1} (Fig. 2c,d).

Seasonal variation in the daily mean seawater values were high (Fig. 2, Table 1). Daily mean *f*CO₂ values were on average 67 μatm higher in the two warmest months (January–February: 406 μatm) compared to the two coolest months (July–August: 339 μatm). Seawater temperature varied seasonally by 5.3 °C (27.0 °C versus 21.7 °C), whereas salinity averaged 35.5 with overall relatively minor systematic seasonal variation, but brief downward spikes that occurred mostly in the wettest summer seasons (Fig. 2d).

Diel variation in *f*CO₂ often exceeded 50 μatm and at times > 100 μatm, particularly in summer (Fig. 3a). Diel *f*CO₂ ranges averaged 37 ± 0.71 μatm across the six warmest months (November–April), and 24 ± 0.50 μatm in the six cooler months. Diel ranges in seawater temperature averaged 0.56 ± 0.01 °C and 0.41 ± 0.01 °C in summer and winter, respectively. Overall, mean *f*CO₂ values were highest during summer nights (midnight to 4 am) averaging 389 μatm (376 μatm in winter nights; Fig. 3b). They were lowest on winter afternoons (noon to 4 pm), averaging 333 ± 0.58 μatm (341 μatm on summer afternoons). Hence mean diel variation in *f*CO₂ was about 30–50% of the ranges of the mean seasonal variation, and was attributable to changes in temperature and reef metabolism (photosynthetic CO₂ uptake and night respiration).

To determine the relative strength of the estimated decadal trends and the role of co-varying environmental factors, non-hierarchical GLMs and partial dependence plots were used on monthly averaged data. The decadal upward trends in atmospheric CO₂, seawater temperature and salinity were significant, with very strong effects of month (representing seasonal variation) for temperature and much weaker effects of month for atmospheric CO₂ and for salinity (Table 2a). Seawater *f*CO₂ also displayed a strongly significant decadal upward linear trend, and additionally it varied with temperature, salinity, and month (Fig. 4, Tables 1, 2b). Thus, after adjusting for the variation in seawater temperature, salinity and months using the GLMs, the decadal trend in *f*CO₂ remained strong and significant (1.792 ± 0.3612 μatm year^{–1}, *t* = 4.960, *P* < 0.001). The decadal trend was similarly strong for *f*CO₂-T, i.e., after temperature normalisation to remove the influence of seawater temperature on the *f*CO₂ trend following Takahashi et al.³⁹, the decadal *f*CO₂-T trend was 1.828 ± 0.628 μatm year^{–1} (*t* = 5.009, *P* < 0.001). The strongest predictor for variation in both *f*CO₂ and *f*CO₂-T at GBRWIS was the decadal linear trend (Table 2b).

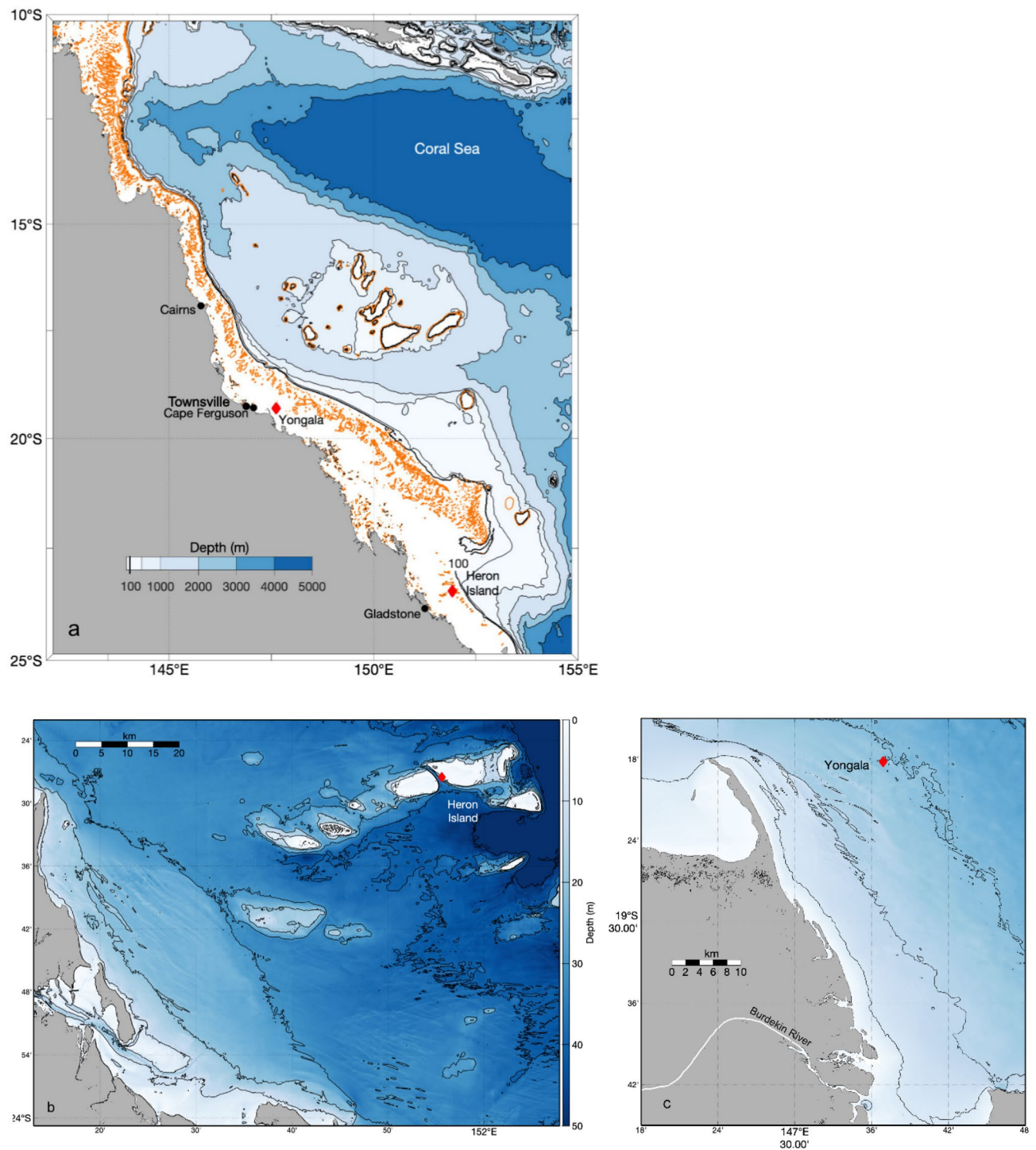


Figure 1. Map of the Great Barrier Reef (orange lines: coral reefs) and the atmospheric gas reference station Cape Ferguson (a), and of the oceanographic stations GBRWIS (b) and NRSYON (c). The blue shading indicates bathymetry, with the thick black line showing the 100-m bathymetry line. The figures were generated using bathymetry data from the 30 m high resolution depth model for the Great Barrier Reef of R. J. Beaman downloaded from Geoscience Australia (<https://ecat.ga.gov.au/geonetwork/srv/eng/catalog.search#/metadata/115066>), and plotted using Matlab version R2020a software with the M_Map mapping package (version 1.4 m, created by R. Pawlowicz, <https://www.eoas.ubc.ca/~rich/map.html>).

NRSYON station. Atmospheric CO₂ concentrations at the Cape Ferguson greenhouse gas station increased near-monotonically from 353 ppm in June 1991 to 408 ppm in July 2019. Between 2009 and 2019, the mean rate of increase was 2.37 ± 0.03 ppm CO₂ year⁻¹ (Fig. 5a, Table 1).

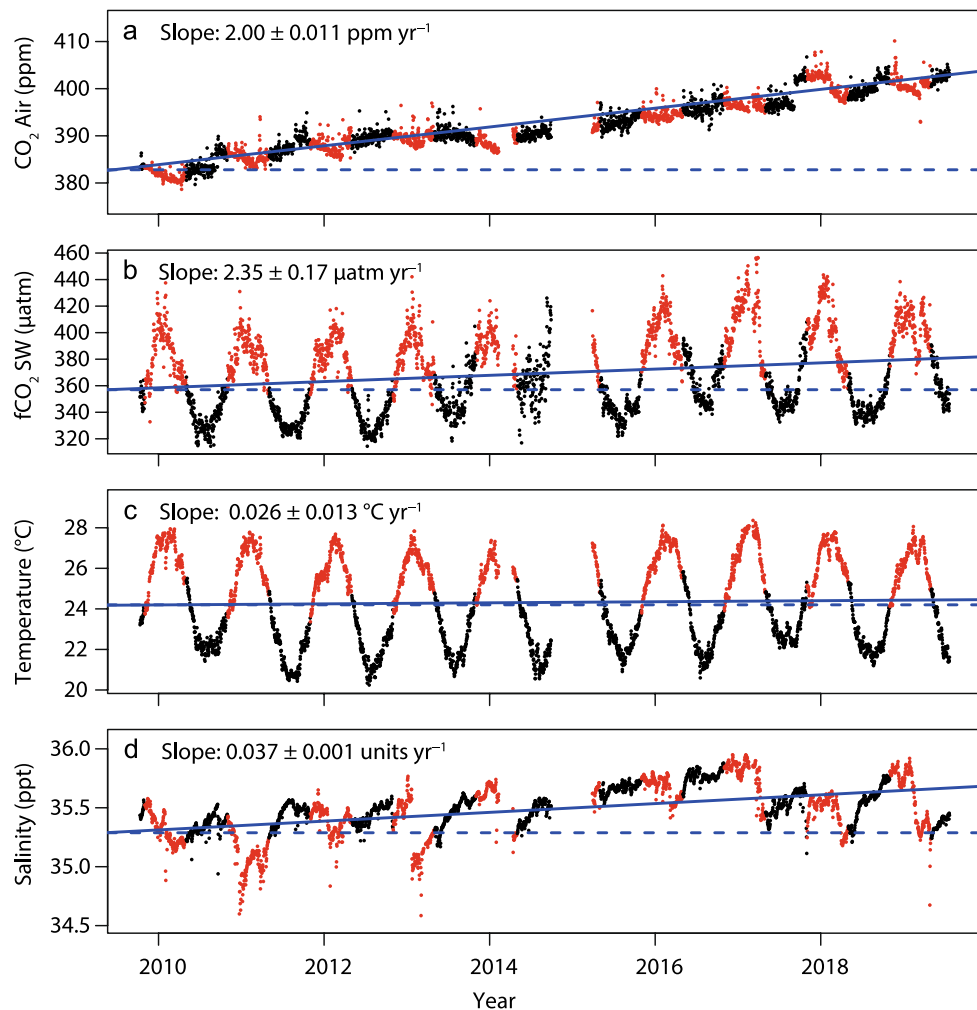


Figure 2. GBRWIS: observed daily averaged data of (a) atmospheric CO₂ (ppm), (b) fugacity of CO₂ in seawater (*f*CO₂, µatm), (c) seawater temperature (°C) and (d) salinity. Red points indicate austral summer (November–April, inclusively), black indicate winter (May–October). Solid blue lines show the linear trends, dashed blue lines are horizontal guides. Also shown are the slopes ± 1 standard error.

The NRSYON seawater samples showed no overall significant differences across the four sampling depths (Fig. 5b,c), despite some individual samples showing depth differences that suggested occasional weak stratification especially in summer. Based on the absence of systematic depths differences, all data were averaged across depths within sampling months for the following analyses ($N = 107$).

NRSYON data displayed strong seasonal variation (Fig. 5b–g; Table 1). *f*CO₂ values ranged over 113 µatm, being on average 58 µatm higher in the warmest vs the coolest months (402.5 µatm vs 344.3 µatm). Ω_{ar} was also higher (3.76 vs 3.45). Mean seawater temperatures were 6 °C higher during the two warmest compared to the two coolest months (January/February: 28.7 °C, July/August: 22.7 °C), while salinity and pH_T were lower in the warmest versus the coolest months. Mean concentrations of dissolved inorganic nitrogen, total phosphorus and silicate were relatively low, however variation was high (Table 1).

Superimposed over the seasonal variation, many of the NRSYON data showed strong decadal trends (Fig. 5b–g). Measured DIC increased by 3.09 ± 0.67 µmol kg^{−1} SW year^{−1}. Calculated *f*CO₂ increased on average by 2.00 ± 0.92 µatm year^{−1}, at a rate that was not significantly different from that of the atmospheric CO₂ trend at nearby Cape Ferguson (2.37 ± 0.03 ppm CO₂ year^{−1}; GLS, $P = 0.69$). Calculated pH_T declined by 0.00156 ± 0.0009 units, and Ω_{ar} by 0.0044 ± 0.0048 units year^{−1}. Seawater temperature and salinity also increased over the decade (Fig. 5f–g).

The statistical analyses showed the decadal upward trends in temperature and salinity to be significant ($P < 0.05$), albeit only marginally so for temperature (non-hierarchical GLM analyses, also accounting for seasonal variation; Table 3). For DIC, *f*CO₂, pH and Ω_{ar} , but not for A_T, the decadal trends remained significant after accounting for seasonal variation (months) and changes in temperature, salinity and nutrients (dissolved inorganic nitrogen, phosphorus and silicate) (Fig. 6, Supplementary Fig. S1a,b, Table 3). For DIC and A_T, the strongest predictor was salinity, with weaker but significant effects from DIN and month. Calculated *f*CO₂, pH, Ω_{ar} and the Revelle factor were most strongly related to temperature, followed by the decadal trend and silicate. After normalisation to mean salinity values following Friis et al.⁴⁰, the decadal trend in DIC-S became weaker

	Total mean	Min	Max	Range	Winter mean	5%	95%	Summer mean	5%	95%	Decadal trend (units year ⁻¹ ± SE)
Atmospheric CO₂											
Cape Ferguson Air CO ₂ (ppm)	395.4	384.5	407.8	23.3	394.7	385.7	405.6	395.3	385.3	406.7	2.367 ± 0.0320
Mauna Loa Air CO ₂ (ppm)	398.4	384.4	414.7	30.2	398.6	387.5	410.7	397.3	387.9	407.2	2.389 ± 0.0147
GBRWIS											
Air CO ₂ (ppm)	392.4	378.7	410.1	31.4	392.4	383.0	400.8	391.4	381.1	402.4	2.010 ± 0.0634
<i>f</i> CO ₂ (μatm)	369.8	314.4	456.5	142.1	339.4	321.2	364.5	405.9	380.7	435.9	1.792 ± 0.3612
Temperature (°C)	24.32	20.24	28.36	8.13	21.65	20.65	22.65	27.04	25.93	27.84	0.0429 ± 0.0127
Salinity	35.49	34.33	35.95	1.62	35.54	35.35	35.77	35.47	34.95	35.86	0.0387 ± 0.0051
NRSYON											
DIC (μmol kg ⁻¹ SW)	1981	1914	2017	103	1991	1975	2005	1966	1922	1990	0.802 ± 0.239
A _T (μmol kg ⁻¹ SW)	2300	2224	2338	113	2304	2273	2324	2293	2237	2319	0.0976 ± 0.256
pH _T *	8.062	8.003	8.113	0.111	8.096	8.082	8.110	8.035	8.019	8.049	-0.0012 ± 0.0004
Ω _{ar} *	3.586	3.230	3.921	0.691	3.455	3.253	3.589	3.735	3.604	3.842	-0.0073 ± 0.0024
<i>f</i> CO ₂ (μatm)*	377.1	329.5	442.5	113.0	344.3	331.2	356.2	402.5	388.0	421.0	1.244 ± 0.377
Revelle factor*	9.27	8.90	9.71	0.80	9.42	9.24	9.67	9.09	8.99	9.21	0.011 ± 0.0033
Temperature (°C)	25.77	21.16	29.65	8.49	22.67	21.18	23.80	28.66	27.80	29.44	0.064 ± 0.0269
Salinity	35.29	34.02	36.05	2.02	35.34	34.83	35.60	35.14	34.15	35.62	0.0587 ± 0.0099
DIN (μmol kg ⁻¹)	0.111	bdl	1.137	1.137	0.123	bdl	0.443	0.106	bdl	0.373	ns
Phosphate (μmol L ⁻¹)	0.063	bdl	0.24	0.24	0.077	0.02	0.16	0.044	bdl	0.123	ns
Si (μmol L ⁻¹)	1.29	0.10	3.80	3.7	1.11	0.6	1.73	1.46	0.37	3.09	ns

Table 1. Atmospheric CO₂ and surface seawater conditions at the two Great Barrier Reef stations averaged over the years 2009–2019. Total means and ranges: GBRWIS: 3036 daily mean instrument-measured values, NRSYON: 109 monthly mean (depth-averaged) seawater samples. Atmospheric CO₂ concentrations from Cape Ferguson (<https://gaw.kishou.go.jp/search/station#CFA>) and from Mauna Loa (sourced from www.esrl.noaa.gov/gmd/ccgg/trends/), and the means and percentiles for the two warmest (January and February) and coldest months (July and August). Also tabled are the estimated slopes (± standard errors) for the decadal trends derived from non-hierarchical GLM analyses of monthly averaged data (the seawater carbonate data corrected for variation with seasons, temperature, salinity and nutrients; Figs. 4, 6; Tables 2, 3; ns = not significant). *Variables calculated from DIC and A_T for in situ temperature, pressure, nutrients and salinity, using CO2Sys⁶¹ with pK₁ and pK₂ dissociation constants of Dickson and Millero⁶² and KH₂SO₄ dissociation constants of Dickson⁶³, total borate of Uppstrom⁶¹ and the equilibrium constant for HF from Perez and Fraga⁶². pH_T pH at total scale, Ω_{ar}, aragonite saturation state, *f*CO₂ fugacity of seawater CO₂, bdl below detection limit. Measurements of DIC and A_T used for CO₂ system calculations were only available for NRSYON.

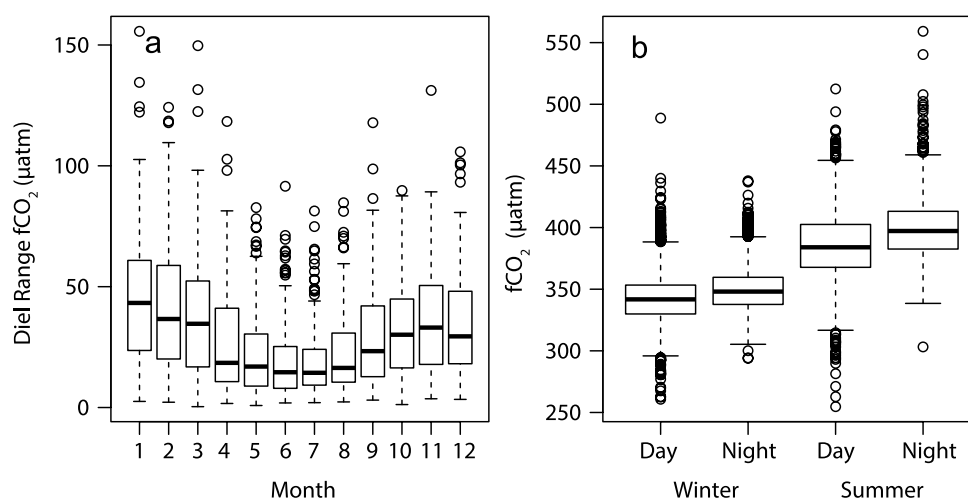


Figure 3. Diel and seasonal variation in *f*CO₂ at GBRWIS: (a) changes in diel ranges of seawater *f*CO₂ (μatm) across months. (b) Mean *f*CO₂ at day (12:00–16:00 pm) and night (0:00–4:00 am), and in winter (May–Oct) and summer (November–April). Box-whisker plots, with horizontal bars indicating the median, boxes are the 25th–75th percentiles, dashed whiskers are 1.5 interquartile ranges, and circles represent outliers (only data from complete years are included).

(a)	df	Atmospheric CO ₂		Temperature		Salinity	
		F	P	F	P	F	P
Decadal trend	1	1881.1	<0.001	11.42	0.001	58.43	<0.001
Month	11	2.789	0.0034	267.5	<0.001	3.527	<0.001
(b)	df	<i>f</i> CO ₂		<i>f</i> CO ₂ -T			
		F	P	F	P		
Decadal trend	1	24.6	<0.001	25.10	<0.001		
Month	11	4.148	<0.001	3.818	<0.001		
Temperature	1	19.9	<0.001	5.283	0.024		
Salinity	1	4.399	0.039	2.959	0.089		

Table 2. GBRWIS: factors related to the observed variation in (a) atmospheric CO₂, seawater temperature and salinity, and in (b) *f*CO₂ (Fig. 4) and *f*CO₂ normalised to the local mean temperature (24.34 °C)³⁹. Non-hierarchical generalized linear model (GLM) analysis of monthly mean data (N = 109 months).

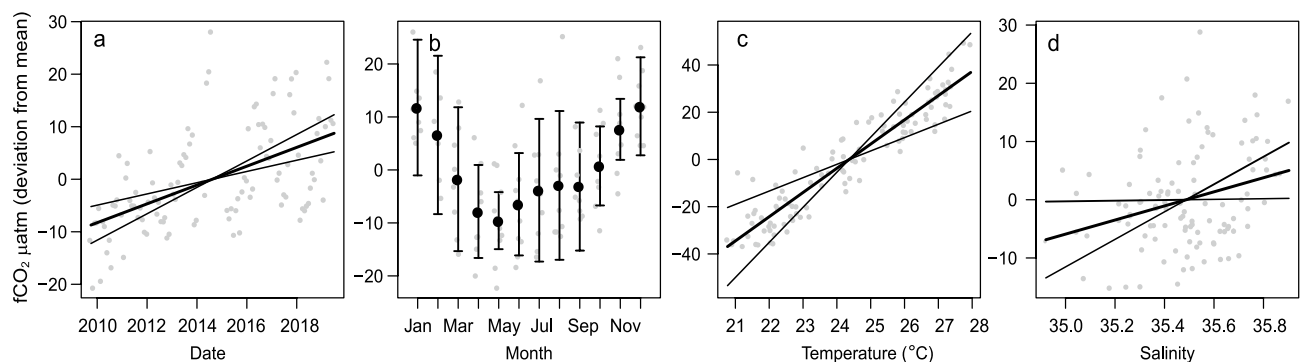


Figure 4. GBRWIS: partial dependency plots showing the changes in estimated monthly mean seawater *f*CO₂ throughout the decade (a), across months (b), with temperature (c) and with salinity (d), while controlling for all other factors (Table 1). y-axis values represent the difference of the predicted *f*CO₂ response from the mean value (369.8 μatm *f*CO₂), solid thick lines show the model estimates, i.e., decadal trends (date), seasonal variation (months) and changes related to temperature and salinity. Thin lines and error bars indicate 2 standard errors of the estimates, and grey dots the residuals. Partial dependency plots display model estimates to any predictor value while holding all other predictors constant at their mean (or categorical) values.

and for A_T-S it became insignificant (Supplementary Table S1, Fig. 1C). However, salinity remained a significant predictor, suggesting slight over-compensation by the normalisation. Normalisation of *f*CO₂ to mean temperature also showed that the decadal trend remained significant at $1.259 \mu\text{atm year}^{-1} \pm 0.3696$, $P < 0.001$ (also accounting for changes in salinity, nutrients, and months; Supplementary Table S1). For the nutrients, silicate was significantly associated with all carbon variables except for A_T. DIN was significantly associated with DIC and DIC-S and marginally with A_T and A_T-S. Phosphate was only weakly associated with DIC and DIC-S.

Discussion

Long-term trends. Two long-term monitoring stations in the GBR, 650 km apart in very different environmental settings, display significant upward trends in their near-surface seawater carbon chemistry. The two GBR stations show very similar rates of *f*CO₂ increase, at $\sim 2.0 \pm 0.3 \mu\text{atm year}^{-1}$, additional to substantial seasonal and moderate diel variation. These trends are quite similar to rates documented for open ocean regions^{12,13,19,41,42}. For example, pCO₂ at the pelagic Hawaiian Station ALOHA increased at $2.5 \pm 0.1 \text{ ppm year}^{-1}$ between 1988 and 2002, albeit with much weaker ($\sim 20 \mu\text{atm}$) seasonal variation⁴³. Similarly, two oceanic long-term monitoring stations, WHOTS in the subtropical North Pacific and Stratus in the South Pacific gyre, show pCO₂ trends of 1.9 ± 0.3 and $1.6 \pm 0.3 \mu\text{atm year}^{-1}$, respectively. At the North Atlantic Ocean ESTOC site (offshore from the Canary Islands), *f*CO₂ increased by $1.55 \pm 0.43 \mu\text{atm year}^{-1}$ between 1995 and 2004, which the authors attributed to atmospheric forcing together with large-scale oceanic and climatic features⁴⁴. For pH, the NRSYON trend ($-0.0016 \pm 0.0009 \text{ pH units year}^{-1}$) was also similar to other published offshore and pelagic long-term seawater carbon monitoring stations (-0.0016 to $-0.0018 \text{ pH units year}^{-1}$), in agreement with those expected under air-sea CO₂ equilibration (-0.0016 to $-0.0018 \text{ pH units year}^{-1}$)^{19,44}.

The similarity in the decadal CO₂ trends at both GBR stations to the atmospheric CO₂ changes suggests these CO₂ trends are largely determined by atmospheric forcing, as also found at the open ocean sites. This assumes there has been no significant reorganisation of the biological and physical processes influencing surface water carbon chemistry over the decade of measurements at our two sites. NRSYON is strongly influenced by coastal processes and is close to a major river outflow, but $\sim 100 \text{ km}$ from the shelf edge, whereas the GBRWIS site is

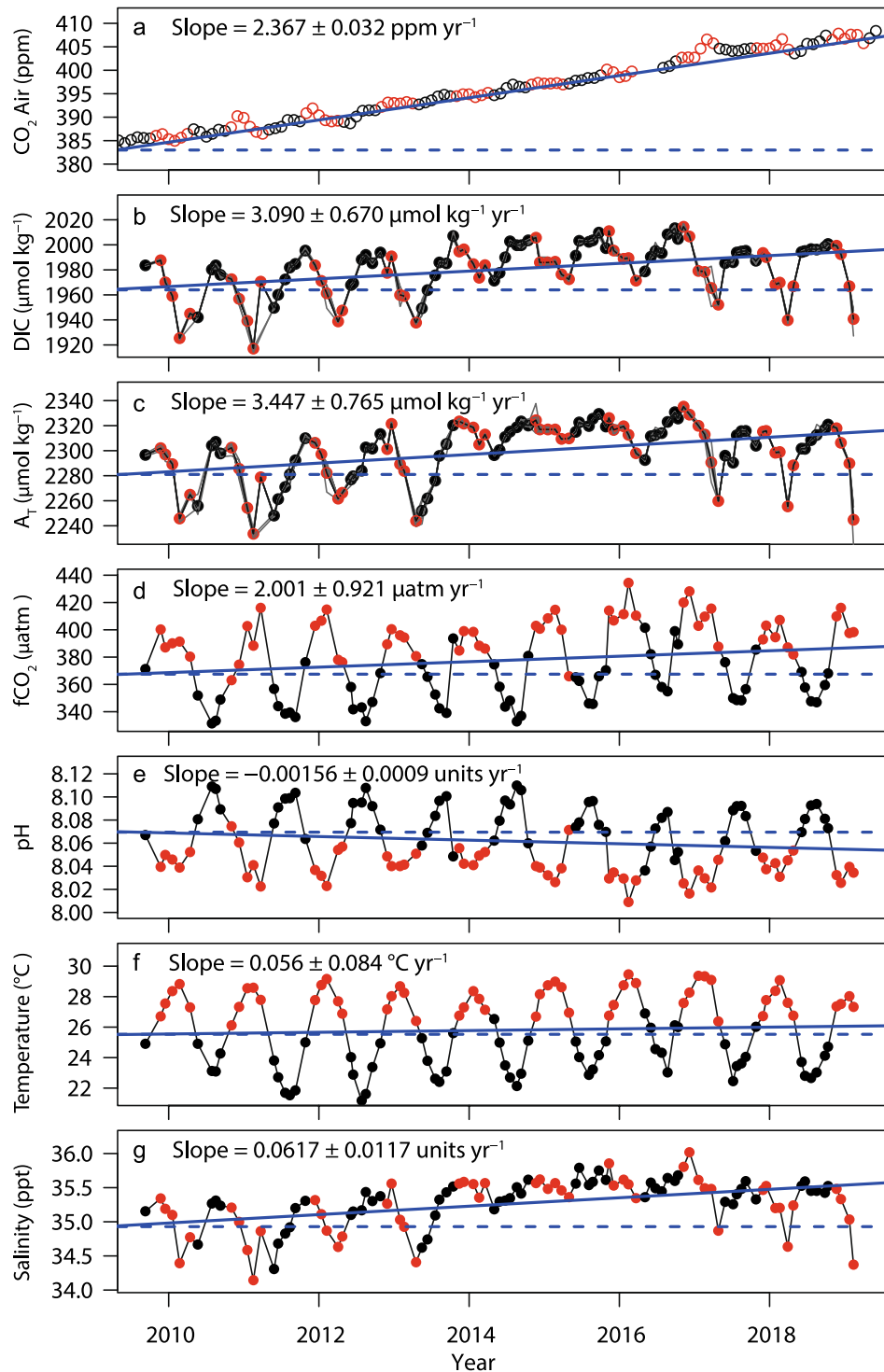


Figure 5. NRSYON: observed atmospheric CO_2 and seawater carbon chemistry values from 2009 to 2019. (a) Atmospheric CO_2 concentrations at the Cape Ferguson greenhouse gas monitoring station. (b) To (g): dissolved inorganic carbon (DIC), total alkalinity (A_T), temperature and salinity (all measured); $f\text{CO}_2$ and pH (calculated). Points are depth-averaged monthly seawater samples; red points indicate summer (November–April); black are winter (May–October). In (b,c), the thin grey lines show the four sampling depths (0, 10, 20, 26 m). Straight solid blue lines are the linear trends, dashed blue lines are horizontal guides. Also shown are slopes of daily averaged data, unadjusted for changes in temperature and salinity.

(a)	df	Temperature		Salinity		DIC		A _T	
		F	P	F	P	F	P	F	P
Decadal trend	1	5.678	0.037	34.97	<0.001	11.21	0.001	0.145	0.704
Month	11	86.53	<0.001	4.947	<0.001	4.517	<0.001	3.859	<0.001
Temperature	1					2.119	0.149	6.673	0.011
Salinity	1					361.2	<0.001	514.1	<0.001
DIN	1					7.116	0.009	4.151	0.045
Phosphate	1					4.625	0.034	1.118	0.293
Silicate	1					0.971	0.327	4.671	0.033
(b)	df	fCO ₂		pH		Ω _{ar}		Revelle Factor	
		F	P	F	P	F	P	F	P
Decadal trend	1	10.92	0.001	10.80	0.002	8.994	0.004	11.30	0.001
Month	11	1.744	0.076	1.829	0.061	1.812	0.063	1.865	0.055
Temperature	1	50.15	<0.001	55.64	0.000	47.61	<0.001	31.78	<0.001
Salinity	1	6.161	0.015	0.272	0.604	25.08	<0.001	4.973	0.028
DIN	1	1.175	0.281	0.560	0.456	0.154	0.696	0.825	0.366
Phosphate	1	2.416	0.124	1.885	0.173	1.016	0.316	2.362	0.128
Silicate	1	11.16	0.001	11.49	0.001	12.63	0.001	10.43	0.002

Table 3. NRSYON: environmental and temporal factors associated with changes in seawater carbon chemistry variables. Non-hierarchical generalized linear model (GLM) analysis of monthly depth-averaged data (N = 107 months). Tabled are the decadal trend and variation due to seasons (Month) for seawater temperature and salinity, and the decadal trend, seasonality and environmental predictors (seawater temperature, salinity and nutrients) for selected carbon chemistry variables: measured DIC and A_T, and calculated fCO₂, pH, and the Revelle factor (Fig. 6, Supplementary Figure S1a,b). Salinity and Temperature normalised values are shown in Supplementary Table S1 and Supplementary Fig. S1c.

located next to a coral reef but 68 km off the coast and within 20 km of the shelf edge and the open-ocean waters of the Coral Sea. The buffering capacity by the carbonate sediments that cover most of the seafloor in the GBR was clearly insufficient to prevent the rapid acidification of the water body near these two stations for the 10-year duration of the measurements. Based on this similarity we suggest the observed decadal trends are unlikely to be exclusive to the two sites and reflect the trend over a broad region of the central and South GBR.

Seawater temperature and salinity also showed significant upward trends over time at the two GBR stations, consistent with regional oceanographic and climatological features. The physical oceanography of the GBR is dominated by the inflow of Coral Sea water which bifurcates offshore between about 15 and 20° S, leading to predominantly southward-bound offshore currents, and northward bound inshore counter currents in the southern and central GBR⁴⁵. Sea surface temperatures are increasing over the whole GBR region with changes in Coral Sea source waters of 0.1–0.2 °C per decade⁴⁶. The variable salinity trend is consistent with temporal and regional variation in monsoonal rainfall and river flows, and also reflects slight salinity increases in Coral Sea waters⁴¹. The climate of Northeast Australia is additionally influenced by ENSO cycles, which typically have a 2–7 years duration. At both sites, El Niño (La Niña) events were associated with anomalously high (low) salinities compared to the long-term trend. La Niña conditions with increased rainfall were present from mid-2010 until about March 2012⁴⁷. By mid-2014, a moderate to strong El Niño with low rainfall emerged and persisted until mid-2016, followed by neutral conditions⁴⁸. At GBRWIS, the downward spikes in salinity coincided with large rainfall events in 2010/11, 2013 and 2019. The coastal NRSYON site had a greater variability and trend in salinity than GBRWIS, likely attributable to its greater exposure to coastal inflows, and the influence of net evaporation or precipitation in shallower water. The Burdekin River had large discharges and northward plume flows in the summers of 2007/08 to 2011/12 and 2018 and 2019 with seasonally low salinities at NRSYON, and low discharges from 2012/13 to 2017/18 that corresponded with higher salinities in the summer wet season at the site.

Both temperature and salinity play an important role in determining changes in the seawater carbon chemistry, directly as temperature determines the solubility of CO₂ and salinity affects A_T and DIC, and indirectly as biotic metabolism accelerates with temperature. Total alkalinity at NRSYON was strongly related to salinity, with prolonged reductions in years with high river flows. The decadal increase in temperature and salinity, and resulting changes in A_T and DIC at NRSYON will cause an increase in fCO₂ and decrease in pH and Ω_{ar}. However, the decadal upward trend in fCO₂ remained significant after normalisation to a mean temperature at both sites, and also after statistically adjusting for the effects of rising temperatures and salinity. This again confirms the strong role of atmospheric forcing in driving the trends in seawater CO₂ chemistry over the duration of our study. In addition to the decadal trends at both sites, there is substantial seasonal variation in the carbon chemistry due to seasonal temperature fluctuations and biological activity. Longer term observations through multiple ENSO events are needed to resolve the combined influences of local and large-scale controls on the rate of ocean acidification in the GBR more clearly.

We hindcasted fCO₂ conditions in the GBR over the last 60 years, when the longest existing greenhouse gas records commenced at Mauna Loa, Hawaii (since 1958; www.esrl.noaa.gov/gmd/ccgg/trends/). The decadal

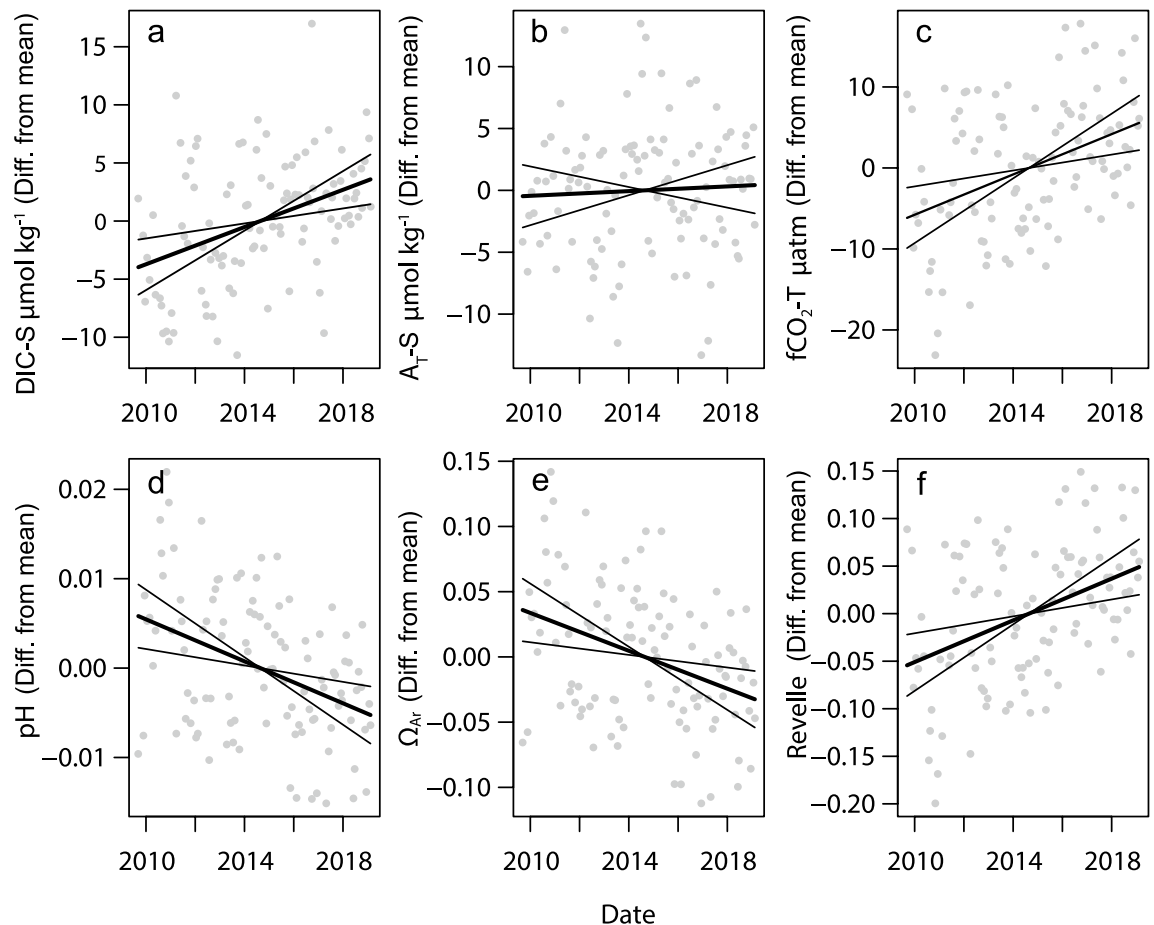


Figure 6. NRSYON: partial dependence plots showing the changes in estimated depth-averaged DIC and A_T (a,b), and the other carbon chemistry variables (c–f) related to decadal trends (date). Trend estimates of are controlled for changes across months and in temperature, salinity and nutrients (Table 3, Supplementary Fig. S1). Values on the y-axis represent differences from the mean values (Table 1). Solid thick lines are the model estimates, thin lines are 2 standard errors of the estimates, and grey dots the residuals.

trends in the atmospheric CO_2 time series from both GBRWIS and Cape Ferguson closely matched the winter values of Mauna Loa (Fig. 7). From 2009 to 2019, atmospheric CO_2 at Mauna Loa increased by $\sim 2.2 \pm 0.01$ ppm year $^{-1}$ (NOAA 2019), at GBRWIS by 2.00 ± 0.011 ppm year $^{-1}$, and at Cape Ferguson by 2.37 ± 0.03 ppm year $^{-1}$ (Figs. 2, 5, 7). The slopes of seawater $f\text{CO}_2$ at GBRWIS and NRSYON over the same time period were also statistically similar, increasing at 2.35 ± 0.17 and 2.00 ± 0.92 $\mu\text{atm year}^{-1}$ respectively, with an offset by ~ 26 ppm for GBRWIS and ~ 13 μatm for NRSYON compared with their atmospheric CO_2 values. Our data suggest that today's $f\text{CO}_2$ minima are likely above the maxima these two sites would have experienced in the early 1960s (Fig. 7). At NRSYON, the current (2009–2019) average range of winter to summer $f\text{CO}_2$ is 344.3–402.5 μatm (Table 1) and extrapolation back to 1958 gives a range of about 261–319 μatm , i.e., the current envelope is outside than from 50 years ago. At GBRWIS, only $f\text{CO}_2$ and no other systematically sampled carbon chemistry data are available over the 10-year sampling period, however the present day average winter–summer range of $f\text{CO}_2$ (339.4–405.9 μatm) is also probably outside the average seasonal range in the early 1960s of 333–250 μatm (Fig. 7). Our hindcasts assume that $f\text{CO}_2$ at the GBR sites have tracked the atmospheric increase at Mauna Loa, and that GBR seawater A_T has remained similar since 1958. The observed long-term changes in temperature and salinity are too small to alter this result. The increasing Revelle Factor with time due to surface water CO_2 uptake has likely amplified rather than diminished the seasonal signal today relative to that of the 1960s. Factors such as a widespread decline in net calcification on the reef since the 1960s may have altered the $f\text{CO}_2$ ranges by changing A_T and DIC concentrations, however few data from before 2009 exist to determine if this is has occurred. Therefore, despite the highly dynamic nature of both stations, their $f\text{CO}_2$ concentrations are now likely about 28% higher than in the 1960s and outside the envelope that both locations would have experienced only 60 years ago. These findings match those from open ocean subtropical and subarctic moorings, where today's surface seawater chemistry conditions are also largely or entirely outside of the bounds of preindustrial values throughout the year²⁰.

At NRSYON, average Ω_{Ar} would likely have been ~ 0.33 units higher at ~ 3.92 in the 1960s than it is today at 3.59. Similarly, today's pH at NRSYON averages 8.062, and has likely dropped by ~ 0.07 units, so pH may have averaged around 8.13 at this site in the early 1960s. These estimates add further confidence to earlier

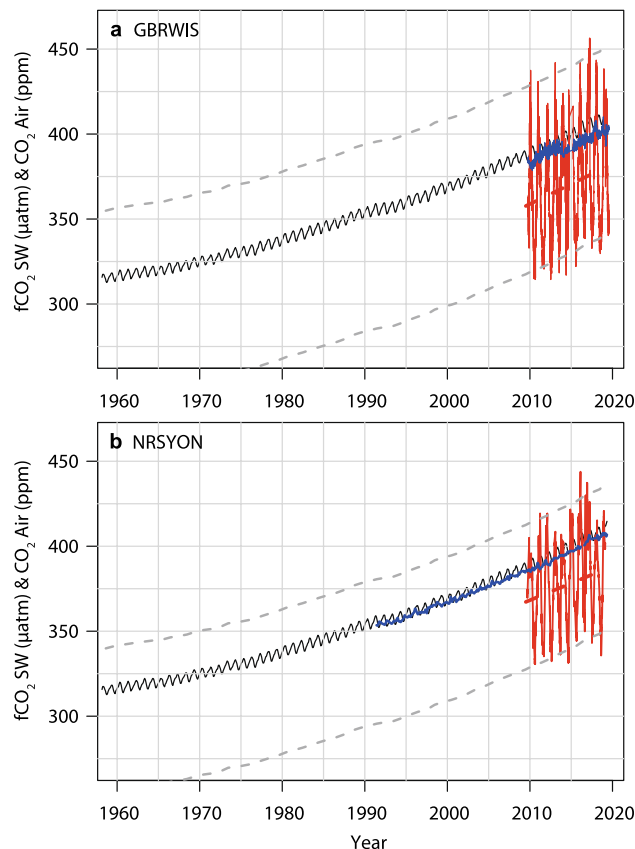


Figure 7. Observed changes in atmospheric CO₂ concentrations at Mauna Loa monitoring station¹ (solid black line), and at GBRWIS since 2009 (a, blue line) and Cape Ferguson since 1991 (b, blue line; sourced from <https://gaw.kishou.go.jp/search/station#CFA>). Red lines: observed seawater $f\text{CO}_2$ and their linear fits at GBRWIS (a) and NRSYON (b). Grey dashed lines are back-projections of observed seawater $f\text{CO}_2$ envelopes. Also shown are slopes of the data prior to correction for variation in temperature, salinity, seasonality, and nutrients.

reconstructions of changes in the seawater carbon chemistry in the Australian region from salinity, atmospheric CO₂ and temperature data³⁷, which suggested an Australia-wide mean decline in Ω_{ar} by 0.48 units and in pH by 0.09 units since 1870, in response to increasing oceanic uptake of atmospheric CO₂.

Variation in seawater $f\text{CO}_2$, and the role of environmental drivers. Surface seawater carbon concentrations showed substantial seasonal variation compared to the atmospheric CO₂, likely attributable to seasonal changes in temperature and salinity, to benthic and pelagic production, as well as to differences in source water bodies²⁵. However, observed ranges in $f\text{CO}_2$ at the two GBR stations were moderate compared to many other coastal stations¹⁵. Total ranges were 143 μatm at GBRWIS and 113 μatm at NRSYON, while seasonal ranges averaged 67 μatm at GBRWIS and 58 μatm at NRSYON. Diel ranges were only assessed at GBRWIS, where they averaged 40 μatm in summer and 24 μatm in winter. In comparison, at moorings around O’ahu, Hawai’i (2008–2016), the coastal pCO₂ station in Kaneohe Bay behind a strongly calcifying shallow coral reef with slow turnover and strong terrestrial influences displayed a maximum CO₂ range of 950 μatm pCO₂, while two other Hawai’ian stations with less calcification exhibited surface water pCO₂ variability ranging from 198 to 240 μatm²¹. Mean diurnal variability at Kaneohe Bay at 192 μatm was also 4–5 times greater than at GBRWIS. In Hawai’i as in the GBR, the seasonal seawater CO₂ dynamics were opposite to those in the atmosphere, with highest seawater CO₂ values in summer and lowest values in winter.

Seasonal variation was greater on the mid-shelf GBRWIS than at the coastal NRSYON (Table 2), and there was a larger mean gradient between seawater and atmospheric CO₂ at GBRWIS compared to NRSYON (26 vs 13 ppm). A comparison between the two stations needs to be done with caution: at GBRWIS the $f\text{CO}_2$ was directly measured by instruments day and night and in all weather conditions but only at about 0.5 m water depth. At NRSYON, $f\text{CO}_2$ was calculated from seawater DIC and A_T sampled at four depths through the 26 m water column and samples were only collected during daylight and not in very windy conditions. The subtropical GBRWIS also experiences slightly greater seasonal variation in daylength and temperature compared to the tropical NRSYON, likely contributing to its greater variability. Furthermore GBRWIS is located near shallow and actively calcifying coral reefs, while NRSYON is located near a major river in ~26 m deep turbid water without shallow reef structures nearby, and light on the seafloor is low⁴⁹. Consequently, the greater diel fluctuations in $f\text{CO}_2$ in summer than winter at GBRWIS are likely attributable to daytime photosynthesis and night respiration

being greater at the higher temperatures and higher PAR in summer. Considering all these factors, it appears the biological processes on the coral reefs had a greater effect on the local carbon chemistry around GBRWIS than did the more variable salinity, nutrients, sediments and inter-reefal benthos near the Burdekin River at NRSYON.

Seawater nutrient concentrations were positively related to some of the carbon chemistry variables at NRSYON (Supplementary Fig. S1, Table 2). However mean concentrations of dissolved macronutrient are typically low in the tropics except in waters exposed to terrestrial runoff and upwelling, with maximum observed concentrations of phosphate and DIN of 0.24 and 3.8 mmol/kg, respectively (Table 1). River nutrient load data are only available on a yearly basis, so could not be included into the models. However, at this site macronutrient availability did not appear to be a dominant driver of the CO₂ variation on seasonal scales. Overall, concentrations of alkalinity and DIC appear typically lower in Burdekin River discharge waters than in seawater, but they are higher in alkalinity than in DIC⁵⁰. Due to the sparsity of relevant river data and complex hydrodynamics, it remains unclear to what extent rivers affect the alkalinity and inorganic carbon concentrations of GBR ecosystems, either directly through their DIC, alkalinity and nutrient loads, or indirectly via stimulation of biological processes.

Conclusions

We have shown that the carbon chemistry at the two GBR stations has changed rapidly over the last decade, with decadal trends that are consistent across locations. The close tracking of atmospheric and aquatic CO₂ in the GBR suggests that its *f*CO₂ has increased by about 28% since the early 1960s, and that projections of environmental conditions from large scale-carbon cycle models may be used to track changes in the carbon chemistry within the GBR. Our data demonstrate that ocean acidification is not merely a concern for the future of the GBR. Rather, it is already part of its rapidly changing chemical and physical conditions in the Anthropocene. The potential buffering capacity of the surrounding shallow carbonate seafloor in this continental shelf system, and the dynamics caused by the metabolism on its coral reefs and its large rivers, are ineffective in protecting this vast system from ocean acidification based on the ~ 10 years of data we analysed. Seawater *f*CO₂ concentrations in the GBR are now outside the envelope the GBR experienced in the 1960s, and trends are expected to continue and even accelerate throughout this century, due to the increasing Revelle factor^{51,52}.

The implications of this pervasive and rapid change in the concentration of a biologically highly active parameter together with seawater warming, will greatly depend on the location of potential refugia from ocean acidification (if any), and CO₂ emission trajectories. A large number of experimental and in situ studies around volcanic CO₂ seeps have demonstrated the profound implications of ocean acidification for the ecological integrity of coral reefs, including reduced coral calcification, coral biodiversity, coral recruitment, structural complexity, reef accretion rates, and coralline algal cover, and increasing seagrass and macroalgal cover^{8,9,53}. Today's mean Ω_{ar} at NRSYON (3.59, down from a predicted historical mean value of ~ 3.92 in the 1960s) is near the measured tipping point of Ω_{ar} = 3.5 to 3.6 for the GBR, at which level crustose coralline algae and the densities of coral juveniles steeply decline and macroalgal cover increases¹⁰. Eyre, et al.¹⁶ also concluded that the GBR is close to a tipping point when dissolution of carbonate sediments on coral reefs exceeds its generation. Although our data do not support the notion that the carbon chemistry trends at the two sites are modified as yet by net dissolution, longer term records and more measurements in the proximity of benthos are needed to assess this concern. Indeed, the contrasting setting of these stations is valuable in assessing whether or not the GBR is tipping towards conditions of net dissolution. Our study involved only two sites in the south and central GBR, and more data are needed from along its 2300 km length to characterize in greater detail how conditions are changing over time in the broad range of GBR environments (coral reefs, coastal regions including mangroves and estuaries, seagrass meadows and inter-reef regions).

The ongoing support for a global network of *f*CO₂ and carbon chemistry monitoring sites will remain essential to improve predictions of regional seawater conditions and their drivers. Our data complement other studies^{54,55} that show that ocean acidification driven by CO₂ uptake from the atmosphere, together with the rapid increases in global temperatures, is an immediate concern for the persistence of healthy ecosystems in the GBR, rather than a future problem. The data further contribute to the resounding scientific call for immediate and drastic global cuts in CO₂ emissions, combined with strong local management action, to protect coral reefs into the future.

Methods

GBRWIS station. Concentrations of CO₂ in the air (ppm) and surface seawater fugacity of CO₂ (*f*CO₂, μ atm), temperature (°C) and salinity data were sourced from GBRWIS Station, located in 16 m water depth in a channel between Heron and Wistari Reef, near the eastern side of Heron Island, southern GBR on the mid-shelf (– 23.459° S, 151.927° E; Fig. 1a,b). This CO₂ Acidification Mooring surface buoy is part of the Australian Integrated Marine Observing System (IMOS) National Mooring Network, maintained by CSIRO, Hobart^{36,37}, with newly calibrated sensors deployed every ~ 6 months. Surface water and atmospheric CO₂ are measured on either a 2- or 3-h measurement cycle using a Battelle Seaology pCO₂ monitoring system (MAPCO₂), while temperature and salinity of the surface seawater and the equilibrator are measured with a Seabird Scientific SBE16plusV2. The CO₂ measurement uses a bubble equilibrator with an intake at about 0.5 m depth. Air from the equilibrator headspace is circulated through a LI-COR 820 non-dispersive infrared detector (NDIR)^{56,57} and a two-point calibration is automatically conducted before every seawater and air CO₂ measurement, using a zero CO₂ gas (air cycled through a soda lime chamber to remove CO₂, a nafion drier surrounded by silica gel to remove water vapour) and a CO₂-in-air span gas (500–550 μ mol/mol, prepared by the NOAA Earth Systems Research Laboratory and calibrated on the WMO X2007 scale with a standard deviation of 0.06 μ mol/mol; <https://www.esrl.noaa.gov/gmd/ccl/airstandard.html>). Data from 9/10/2009 up to 12/03/2019 were quality-controlled delayed-mode sourced from the Australian Ocean Data Network (<https://portal.aodn.org.au/search>), those to 13/10/2019 were

near-real-time auto quality-controlled data stream. Data were filtered to exclude values flagged as questionable or bad³⁷. For the trend analysis, data points with salinity < 34 PSU were removed (51 of 36,590 points) before creating daily averages. Days with < 7 samples per day were also removed, and the remaining 3242 daily means were averaged to monthly values. Removal of three incomplete months created 109 monthly values.

NRSYON station. Data of total alkalinity and total dissolved organic carbon, nutrients and salinity, were sourced from bottle samples taken at NRSYON, an IMOS National Reference Station at Yongala shipwreck in the central coastal GBR (− 19.305° S, 147.622° E³⁸). NRSYON is located in 26 m water depth, ~ 35 km downstream from the Burdekin River mouth (Fig. 1a,c). The Burdekin River is the largest river entering into the GBR, with annual mean discharges of 7600 GL freshwater, 4.7 million tons of fine sediments, and 9000 tons of particulate nitrogen³³, causing significant intra- and inter-annual variability in salinity, nutrients and turbidity³¹. Samples were collected by AIMS approximately monthly from September 2009 to February 2019. Water was collected in Niskin bottles from 0, 10, 20 and 26 m depth from a small boat, typically during calm conditions and only during day times. For the analysis of total dissolved inorganic carbon concentrations (DIC [$\mu\text{mol kg}^{-1}$ seawater]) and total alkalinity (A_T [$\mu\text{mol kg}^{-1}$ seawater]), samples were drawn from the Niskin bottles into 250 ml Schott bottles using silicone tubing to avoid bubble formation and minimize headspace, preserved with 125 μl of saturated HgCl_2 , stored at room temperature in darkness, and sent to CSIRO Hobart for analysis. Sampling and analysis followed the IMOS National Reference Stations protocol⁵⁸. A_T was measured by open cell titration using a Metrohm Titrando and followed standard operating procedures⁵⁹. DIC was measured by coulometric titration with a SOMMA instrument⁶⁰, and the salinity of the samples was measured by a Seabird conductivity cell associated with the SOMMA (metadata: <https://catalogue-imos.aodn.org.au/geonetwork/srv/eng/metadata.show?uuid=fa93c66e-0e56-7e1d-e043-08114f8c1b76>). Precision and reproducibility for DIC and A_T was estimated from measurements of seawater reference material (Dickson laboratory, Scripps Institution of Oceanography) and of duplicate samples. Carbon data are available through <https://portal.aodn.org.au/search>. Samples for nutrients, including dissolved inorganic nitrogen (NH_4^+ , $\text{NO}_3^-/\text{NO}_2^-$), total silicate and total phosphorus were immediately filtered, and measured spectrophotometrically (metadata: <https://catalogue-imos.aodn.org.au/geonetwork/srv/eng/metadata.show?uuid=fa93c66e-0e56-7e1d-e043-08114f8c1b76>). Temperature data were compiled from data loggers maintained at the station³⁷.

Data of DIC, A_T , temperature, salinity and nutrients were used to compute the other seawater carbon chemistry variables (aragonite saturation state Ω_{ar} , pH at the total scale (pH_T), fugacity and partial pressure of CO_2 ($f\text{CO}_2$ and $p\text{CO}_2$) and the Revelle factor) for observed in situ temperature, pressure, nutrients and salinity, using CO2Sys⁶¹, with pK1 and pK2 dissociation constants of Dickson and Millero⁶² and KHSO_4 dissociation constants of Dickson⁶³, total borate of Uppstrom⁶¹ and the equilibrium constant for HF from Perez and Fraga⁵⁹. For consistency with the GBRWIS $f\text{CO}_2$ data, our reporting of NRSYON CO_2 trends focused on $f\text{CO}_2$. Data points with salinity < 34 PSU were removed for the trend analyses (12 of 432 points). Despite significant depth differences on several days, there were no significant overall differences across the four sampling depths, hence all data were depth-averaged to monthly mean values ($N = 107$). Data on DIC, A_T and $f\text{CO}_2$ data are first presented as observed at in situ temperature and salinity values.

Greenhouse gas monitoring station Cape Ferguson. Regional data of atmospheric CO_2 concentrations as dry mole fractions were sourced from the greenhouse gas monitoring station Cape Ferguson (− 19.2773° S, 147.0587° E, central GBR, ~ 30 km from NRSYON; Fig. 1a). Since May 1991, triplicate discrete air flask samples have been collected monthly at this AIMS jetty, during times when winds were onshore (60°–160°) to minimise land influences. Samples are analysed by CSIRO Oceans and Atmosphere, Climate Science Centre-GASLAB, and provided through the World Data Centre of Greenhouse gases, <https://gaw.kishou.go.jp/search/station#CFA>.

Statistical methods. Statistical analyses were conducted in R v3.4.3 (R Development Core Team⁶⁴), and included the R libraries ‘chron’, ‘doBy’, ‘nlme’, and ‘car’. Analyses of linear trends over the 10-years observation period were based on non-hierarchical generalized linear models (GLMs), with predictors based on date (to estimate the decadal trend), temperature, salinity and month (as chronological factor, to account for other seasonal changes such as irradiance or productivity). Since the focus of this study was on long-term trends rather than short-term fluctuations, the data were averaged within months. For NRSYON, the predictors also included dissolved nutrients (total silicate, total phosphorus, and dissolved inorganic nitrogen as the sum of nitrate and ammonium). Dissolved inorganic nitrogen was fourth-root transformed to approximate normal distribution, while the other variables approximated normal distribution. Estimates of trends in temperature and salinity were based on Gaussian GLMs using the decadal trend (date) and months as the predictors. Model effects were mostly additive, except when indicated. Partial dependence plots were used to show the effects of each predictor on the responses while holding other predictors constant at their mean values (or their respective categorical values)^{10,65,66}. To provide comparison with more traditional non-statistical DIC, A_T and $f\text{CO}_2$ data presentation, the trends analyses were also shown on salinity-normalised DIC and A_T values (DIC-S, A_T -S) following Friis et al.⁴⁰ (mean salinity GBRWIS: 35.49, NRSYON: 35.29), and on temperature normalised $f\text{CO}_2$ ($f\text{CO}_2$ -T) following Takahashi et al.³⁹ (mean seawater temperature GBRWIS: 24.34 °C, NRSYON: 25.81 °C). The results were almost identical to those from the non-hierarchical GLMs as the latter investigate the contribution of each factor while simultaneously controlling for the effects of all other factors included in the models.

We hindcasted the carbonate chemistry of the two data series to 1958, based on the assumption that GBR A_T values, and the slopes and offsets between our GBR data and the data from Mauna Loa since 1958 (www.esrl.noaa.gov/gmd/ccgg/trends/) have remained similar. To do so, we calculated the slopes and offsets for the current

(2009–2019) mean $f\text{CO}_2$ winter minima and summer maxima for both stations and the annual mean Mauna Loa (Table 1), and used these offsets to approximate past $f\text{CO}_2$ envelopes at the stations.

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Author contributions

K.E.F. designed the study, performed the statistical analysis, generated figures and tables, and drafted the manuscript. B.T. co-wrote the manuscript and created the maps. B.T., E.O. and C.N. were instrumental in the data acquisition over 10 years. All authors contributed to interpreting the data and to manuscript revisions, and approved the final submitted manuscript.

Competing interests

The authors declare no competing interests.

Additional information

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