

1 **Earth, Wind, Fire, and Pollution: A Review on Aerosol Nutrient Sources**  
2 **and Impacts on Ocean Biogeochemistry**

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33 A key Earth System science question is the role of atmospheric deposition in supplying  
34 vital nutrients to the phytoplankton that form the base of marine food webs. Industrial and  
35 vehicular pollution, wildfires, volcanoes, biogenic debris, and desert dust all carry  
36 nutrients within their plumes throughout the globe. In remote ocean ecosystems aerosol  
37 deposition represents an essential new source of nutrients for primary production. The  
38 large spatiotemporal variability in aerosols from myriad sources combined with the  
39 differential responses of marine biota to changing fluxes makes understanding of where,  
40 when, and how much nutrients from the atmosphere enter marine ecosystems of crucial  
41 importance. This review brings together existing literature, experimental evidence of  
42 impacts, and new atmospheric nutrient observations compared to atmospheric and ocean  
43 biogeochemistry modeling. We evaluate the contribution and spatiotemporal variability of  
44 nutrient-bearing aerosols from desert dust, wildfire, volcanic, and anthropogenic sources,  
45 including the organic component, deposition fluxes, and oceanic impacts.

## 46 **Introduction**

47 By travelling long distances with the atmospheric flow, aerosols deliver essential nutrients  
48 to remote marine ecosystems (Baker and Jickells, 2017; Doney et al., 2009; Duce et al.,  
49 1991; Jickells et al., 2005; Kanakidou et al., 2012; Mahowald et al., 2018; Zender et al.,  
50 2004). Aerosol transport pathways and distance travelled is determined by a particle's  
51 properties (size, composition, and density), atmospheric conditions (particle uplift,  
52 buoyancy, and the wind speed and direction), and travel altitude. The smaller the particle  
53 size and the higher the altitude of travel, the greater chance for long atmospheric  
54 residence times and for reaching remote marine ecosystems (Baker and Croot, 2010).  
55 The loss of aerosol (gases) from the atmosphere proceeds by two routes: dry (direct)  
56 deposition of the aerosol (gaseous species) and wet deposition within precipitation (Baker  
57 et al., 2007). At ~362 million km<sup>2</sup> the world's oceans cover ~71% of the Earth's surface,  
58 providing a major sink pathway for nutrients and pollutants emitted from land. The  
59 atmospheric lifetime of aerosols ranges from a few days to weeks (Textor et al., 2006),  
60 much less than the mixing-time within-and-between hemispheres. Such short  
61 atmospheric lifetimes combined with the heterogeneous physicochemical nature of  
62 aerosol composition creates large spatial and temporal variations in observed aerosol  
63 nutrient concentration patterns between ocean regions (Figure 1). As understanding of  
64 biogeochemical cycles grows, a strong requirement arises for improved knowledge  
65 regarding how nutrients from different sources are supplied to marine ecosystems.

66 Conceptually, aerosol nutrient sources can be split into three groups: the first is natural in  
67 origin and includes mineral dust, wildfires, volcanoes, and biological particles (Barkley et  
68 al., 2019; Guieu et al., 2005; Jickells et al., 2005; Mahowald et al., 2008; Olgun et al.,

69 2011); the second stems from increasing anthropogenic activity (Ito, 2015; Ito et al., 2019;  
70 Jickells et al., 2017; Krishnamurthy et al., 2009; Luo et al., 2008; Rathod et al., 2020); the  
71 third is the atmospheric transformation of insoluble minerals, of either natural or  
72 anthropogenic origin, into soluble (bioavailable) nutrients during transport by: (1) acids,  
73 (2) organic ligands, and (3) photo-reductive processes (Ito, 2015; Johnson and  
74 Meskhidze, 2013; Longo et al., 2016; Shi et al., 2015; Stockdale et al., 2016). In addition  
75 to directly providing a new nutrient source, anthropogenic activity modifies the generation  
76 of predominately natural aerosols (Ginoux et al., 2012; Hamilton et al., 2018; Mahowald  
77 et al., 2010; Ward et al., 2014), and provides a source of those compounds which work  
78 to liberate a fraction of minerals into more bioavailable nutrient forms (Li et al., 2017;  
79 Meskhidze et al., 2005; Solmon et al., 2009). This coupling of historical human activity  
80 with biogeochemical cycles is highly complex and contains many uncertainties, partly due  
81 to the absence of early historical measurements and on the reliance on proxy records to  
82 describe the historical evolution of natural aerosol fluxes with intensifying human activity.

83 Marine biota are a key component in the biogeochemical cycles that determine the  
84 ocean's capacity to sequester atmospheric CO<sub>2</sub> and support marine ecosystem services  
85 over different timescales. Furthermore, they emit biological gases to the atmosphere  
86 which subsequently oxidize to form organic and sulphate aerosol, therefore creating  
87 climate feedbacks through altering cloud properties, temperatures, and precipitation  
88 rates. By relieving nutrient limitation, large aerosol deposition events have the potential  
89 to modify the natural assemblage due to new resource competition among primary  
90 producers. The resulting change in phytoplankton balance between smaller and larger  
91 sizes (Giovagnetti et al., 2013; Paytan et al., 2009) and between autotrophic and

92 heterotrophic communities (Gazeau et al., 2020; Marañón et al., 2010) determines the  
93 capacity of the ocean to sequester anthropogenic CO<sub>2</sub> following deposition (Guieu et al.,  
94 2014a). The degree to which autotrophs or heterotrophs are stimulated depends on (1)  
95 the physicochemical form in which nutrients are delivered to seawater and (2) the initial  
96 biogeochemical conditions of the water column, including the physical features around  
97 the depth and strength of stratification that determines the initial in situ nutrient limitation,  
98 linked to the supply of macro- or micronutrients from below the thermocline (Guieu et al.,  
99 2014b). Understanding the fertilization potential of aerosol is of particular interest in  
100 surface waters of the Southern Ocean, where phytoplankton net primary productivity  
101 (NPP) is limited by iron supply and biological response to aerosol deposition can regulate  
102 the global carbon cycle (Martin, 1990; Parekh et al., 2004; Tagliabue et al., 2014) and the  
103 redistribution of macronutrients to low latitudes (Sarmiento et al., 2004; Tagliabue et al.,  
104 2009).

105 There has been a recent effort to review the most up-to-date knowledge on the magnitude  
106 of atmospheric deposition fluxes of the main nutrient species to the oceans: Altieri et al.  
107 (2021) reviewed reactive nitrogen; Jickells et al. (2017) updated anthropogenic nitrogen  
108 estimates; Mahowald et al. (2018) reviewed the role of metals as nutrients and pollutants;  
109 Myriokefalitakis et al. (2018) undertook a multi-model comparison for soluble iron; and  
110 Kanakidou (2018) reviewed nitrogen, phosphorus, and iron chemistry and fluxes. From  
111 an oceanic perspective, Boyd and Ellwood (2010) and Tagliabue et al. (2017) reviewed  
112 the marine iron cycle, focusing on the impact of mineral dust deposition. Here, we build  
113 upon this body of literature by reviewing current understanding and uncertainties in major  
114 aerosol nutrient sources (mainly: desert dust, fires, volcanoes, biogenic particles, and

115 anthropogenic activity) and their respective spatiotemporal deposition signatures. While  
116 early studies of the impact of atmospheric iron and phosphorus deposition on ocean  
117 biogeochemistry often solely focused on dust, recent evidence has highlighted the  
118 importance of non-dust sources (Barkley et al., 2019; Ito et al., 2019) - even suggesting  
119 the carbon export efficiency (gram of atmospheric CO<sub>2</sub> sequestered per gram of iron  
120 deposited) of non-dust iron is larger than that of dust-sourced iron (Hamilton et al., 2020a;  
121 Ito et al., 2020b). To further explore the emerging role of (wild)fires in biogeochemical  
122 cycles, we provide a case study of recent megafire activity in Australia during 2019-2020,  
123 including recent observations and new atmospheric and ocean biogeochemical modeling  
124 experiments. The characterization and individual contribution of differing aeolian  
125 pathways to the ocean are important future research areas requiring better constraint to  
126 improve understanding on the impact of aerosol deposition on marine biogeochemical  
127 cycles historically, at present, and in the future, and to better apprehend the fast-evolving  
128 human dimension.

### 129 **Aerosol Nutrients, Observations over Oceans, and Impacts on Biota**

130 Over the last few decades intense research efforts (e.g., SOLAS and GEOTRACES  
131 research programs) have significantly advanced understanding on the oceanic impact of  
132 nutrient-bearing aerosol deposition. That said, field observations are much more  
133 extensive in the Northern Hemisphere due to easier logistics and cheaper cost of  
134 undertaking sea-going campaigns compared to the Southern Hemisphere (Figure 1). As  
135 such, the atmospheric deposition fluxes of nutrients to Southern Hemisphere oceans  
136 remains highly uncertain, with soluble iron the most studied historically. The highly  
137 episodic nature of aeolian dust and wildfire plumes (e.g., the range in daily iron deposition

138 maybe  $\geq 10$  orders of magnitude for some ocean regions; Hamilton et al., 2019) and an  
139 increasing appreciation of the role of atmospheric processing enhancing aerosol nutrient  
140 solubility during transport (Longo et al., 2016; Stockdale et al., 2016) makes data  
141 acquisition of representative field observations a challenge that needs to be addressed.

142 Nitrogen is a core element for sustaining biological systems and functions, but mainly  
143 found in its highly stable elemental form (dinitrogen;  $N_2$ ) which is not bioavailable.  
144 However, through biological nitrogen fixation by diazotrophs  $N_2$  enters the ocean  
145 biosphere and provides the largest source of bioavailable new nitrogen in the open  
146 oceans (Capone et al., 1997; Jickells et al., 2017; La Roche and Breitbarth, 2005). A  
147 smaller amount of reactive nitrogen (nitrogen hereafter) is available which is bioavailable  
148 (Altieri et al., 2021 and references therein) and strongly associated with anthropogenic  
149 sources (Jickells et al., 2017). Owing to the significant amount of literature reviewing  
150 nitrogen supply, further discussion is limited.

151 Phosphorus is another vital nutrient for supporting life. Because heavy particles fall below  
152 the ocean's mixed layer only phosphate may be relevant for marine biota, although  
153 observations remain limited (Baker et al., 2006; Mahowald et al., 2008). Oceanic regions  
154 considered co-limited by phosphate include the western North Atlantic and eastern  
155 Mediterranean Sea (Moore et al., 2013). Additionally, the North Pacific Subtropical Gyre  
156 can be phosphate-limited at times, oscillating with iron limitation (Letelier et al., 2019). In  
157 addition to dust, wildfire and primary biogenic particles are likely to be important sources  
158 of phosphate to much of the open ocean (Barkley et al., 2019; Myriokefalitakis et al.,  
159 2016). Another significant fraction of phosphate, currently unaccounted for in global  
160 modeling, results from the production and use of fertilizer (see below). The diversity of

161 sources partially explains why high phosphate concentrations are observed outside of  
162 major dust plumes from the Northern Hemisphere dust belt and Patagonia (Figure 1).

163 Due to iron's potential to significantly modulate the marine carbon cycle (Cassar et al.,  
164 2007; Martin, 1990) there has been considerable efforts placed on understanding the iron  
165 cycle and its role in supporting both phytoplanktonic NPP in HNLC waters and  
166 diazotrophic uptake of atmospheric N<sub>2</sub> at tropical latitudes. Numerous studies including  
167 in-situ iron-fertilization experiments, ocean surveys, and modeling, have revealed that  
168 iron-limitation of phytoplankton is the primary factor regulating NPP levels in the upwelling  
169 region of the equatorial Pacific, the sub-Arctic Pacific and Atlantic oceans, and in the  
170 Southern Ocean (de Baar et al., 2005; Moore et al., 2013). Here, we explore in detail the  
171 role of iron supply from non-dust sources in marine biogeochemical cycles.

172 The factors that promote mineral dissolution processes, such as atmospheric acidity, the  
173 presence of organic ligands, sunlight, surface-to-volume particle ratios, and ambient  
174 temperature (e.g., Lasaga et al., 1994), represent one of the largest sources of  
175 uncertainty in state-of-the-art model simulations. During long-range atmospheric  
176 transport strong acids, such as sulfuric (H<sub>2</sub>SO<sub>4</sub>) and nitric (HNO<sub>3</sub>), coat phosphorus- and  
177 iron-bearing aerosol and work to liberate them into a soluble (bioavailable) form (Herbert  
178 et al., 2018; Myriokefalitakis et al., 2016; Nenes et al., 2011; Shi et al., 2011). This  
179 phenomena also enriches particles with nitrogen, as already observed for dust (Geng et  
180 al., 2009). Such "well-mixed" deposits produce a favorable cocktail of chemical elements  
181 vital for biota (N, P, Fe) being supplied to the surface ocean. The atmospheric dissolution  
182 of mineral iron is also suggested to be significantly enhanced in the presence of organic  
183 compounds (e.g., with oxalic acid, >75% iron solubility enhancement compared to acid

184 processing alone) (Ito, 2015; Ito and Shi, 2016; Johnson and Meskhidze, 2013;  
185 Myriokefalitakis et al., 2015); see Organic Section for more discussion. Recent modeling  
186 estimates that atmospheric processing creates an additional  $0.56 \pm 0.29$  Tg soluble Fe  
187  $\text{a}^{-1}$  (Myriokefalitakis et al., 2018) and  $0.030$  Gg soluble P  $\text{a}^{-1}$  (Herbert et al., 2018).

188 Post-deposition processes have also been shown to affect atmospheric iron and  
189 phosphorus bioavailability. It is well understood that multiple factors operate in tandem to  
190 determine the fate of atmospheric dissolved iron concentrations ([dFe]), such as: iron  
191 dissolution kinetics, binding ligands, scavenging, biotic uptake (Baker and Croot, 2010;  
192 Boyd and Ellwood, 2010; Bressac and Guieu, 2013; Fishwick et al., 2014). Seasonal  
193 variations in the ocean mixed layer affect dFe concentrations following deposition, with  
194 little change during periods of deep mixed layers and the detrainment of aerosol iron  
195 below the shoaling springtime mixed layer. Depending on the nature and quantity of  
196 dissolved organic matter present at the time of deposition, the same dust/simulated flux  
197 can provoke either iron scavenging (Wagener et al., 2010) or dissolution (Bressac and  
198 Guieu, 2013; Wagener et al., 2008). Post-deposition processes for atmospheric  
199 phosphorus are linked to the amount of iron-oxides present. In abiotic conditions, a  
200 transient release of phosphate is usually observed in seawater, rapidly followed by a  
201 strong concentration decrease due to adsorption onto iron-oxides (Louis et al., 2015,  
202 2018) confirming that interactions between phosphate and iron-oxyhydroxides exert a key  
203 control on phosphate availability in the environment (Chitrakar et al., 2006). Still, the role  
204 of dissolved organic matter via dust-aggregation processes could in some cases, prevent  
205 those interactions (Louis et al., 2015). Cellular iron quotas vary both spatially and  
206 seasonally (Boyd et al., 2012, 2015; Twining et al., 2020; Twining and Baines, 2013) and

207 thus play a crucial role in linking changing aerosol iron supply to the response of  
208 phytoplankton productivity; adding yet another complexity to understanding the  
209 importance of atmospheric deposition. While new modeling efforts start to account for the  
210 scavenging role of dust particles when estimating dFe (Ye and Völker, 2017), more work  
211 is required to account for other aerosol iron phases, which may have distinct sizes,  
212 dissolution kinetics, reactivity and organic phases.

213 Although complex processes are at play in determining the fate of atmospheric nutrients,  
214 some biological response patterns are evidenced by aerosol-addition experiments using  
215 seawater from different regions of the world. Figure 2 presents a compilation of those  
216 data (n=70) reporting a relative change (maximum change shown) in autotroph biomass  
217 following aerosol addition. Although such experiments are not fully representative of the  
218 impact of atmospheric nutrient deposition to the ocean (Guieu et al., 2014b), it is the  
219 parameter for which we have the most data available. Although the average experimental  
220 outcome is in favor of an increase (~+80%) in Chl-a following artificial addition, in many  
221 cases no response or negative impact are reported when (1) aerosol did not relieve  
222 ongoing nutrient limitation (Guieu et al., 2010), (2) when initial metabolic balance toward  
223 heterotrophs was reinforced by the deposition (Gazeau et al., 2020; Marañón et al.,  
224 2010), or (3) when deposition induced a toxic effect (Paytan et al., 2009; Zhang et al.,  
225 2019).

226 These disparities in biological response reveal that the bioavailability of chemical  
227 elements carried by aerosols depends on a complex set of processes. Governing  
228 parameters include the emission source and mixing processes that occurred during  
229 transport, the type of deposition (dry or wet), and the biogeochemical state of the

230 seawater receiving aerosol deposition. Such experiments, which are mostly conducted  
231 over a short period of time, show that the impact (positive or negative) of atmospheric  
232 deposition on the biomass of autotrophs is, when it exists, rapid (a few days). To date,  
233 these experiments have mainly been conducted in oligotrophic environments (Table S1)  
234 and resulted in modest and short-term changes of total biomass. Overall, a mixed (i.e.,  
235 with anthropogenic) aerosol source stimulated Chl-a ~50% more than dust aerosol alone  
236 (dust: +70%, anthropogenic/mixed: +108%). Additional experiments are required to cover  
237 the vast area of the ocean where this type of data remains lacking.

### 238 **The Complex Interplay of Dust and Fire Aerosol Sources**

239 Multi-model estimates of global dust emissions fall between 735 and 8186 Tg a<sup>-1</sup> (median:  
240 2467 Tg a<sup>-1</sup>) with the highest simulated by a model which resolves particle size  
241 distributions above 20µm and up to 63µm (Wu et al., 2020). About 90% of all dust  
242 emissions are located in the Northern Hemisphere dust belt (Ginoux et al., 2012) but are  
243 highly episodic, resulting in the majority (30-to-90%) of dust aerosol deposition to ocean  
244 basins to occur over ~18 days (5%) of days in a year (Mahowald et al., 2009). Dust is a  
245 significant source of mineral nutrients (Figure 3a/b) and how far deposition sustains NPP  
246 is a question asked for many basins (Fung et al., 2000; Guieu et al., 2019; Martin et al.,  
247 1991). Mineral dust phosphorus and iron tends to be relatively insoluble compared to  
248 other sources however (Figure 3a) and, in the case of iron, while dust represents up to  
249 95% of the global atmospheric budget (Mahowald et al., 2009) aluminosilicate and iron-  
250 oxide minerals with extremely insoluble crystalline lattices dominate emissions (<1%  
251 solubility, (Journet et al., 2008)). The composition of soils from which dust originates are  
252 spatially heterogeneous, displaying significantly different characteristics over very small

253 distances (Journet et al., 2014). Soil mineralogy is therefore a critical factor in determining  
254 dust fertilization potential.

255 Fires can also act as a source of bioavailable nutrients to the open ocean (Barkley et al.,  
256 2019; Guieu et al., 2005; Ito et al., 2019; Paris et al., 2010) and are suggested to dominate  
257 the temporal variability in iron deposition fluxes for many basins (Hamilton et al., 2020b).  
258 Two different pathways to aerosol nutrient delivery can be associated with fires (Figure  
259 4). The first is direct emission of aerosols, including the resuspension of terrigenous  
260 particles previously deposited onto vegetation and entrainment of mineral dust particles  
261 from surrounding soils into smoke plumes. Here, emissions are generally driven by the  
262 pyro-dynamics of the fire and thus reflect the intensity of the event and the characteristics  
263 of the fuel being consumed (Reid et al., 2005). The second pathway considers the  
264 saltation and aeolian uplift of the largely bare burnt soil surface resulting from fires.

265 In-situ and lidar remote sensing suggests the presence of mineral dust particles in wildfire  
266 smoke plumes (Kavouras et al., 2012; Nisantzi et al., 2014; Schlosser et al., 2017) and  
267 increased dust emissions from post-fire burnt landscapes have been reported (Dukes et  
268 al., 2018; Jeanneau et al., 2019; Wagenbrenner et al., 2017; Whicker et al., 2006).  
269 However, the total fraction of dust (and hence phosphorus and iron) emissions caused by  
270 fires has not yet been quantified. The average Fe/Al ratio compiled from 30-years' worth  
271 of field campaigns investigating aerosol composition within, mainly tropical, biomass  
272 burning regions (n=17; Tables S3 and S4 and Figure S1) is ~0.6. Fires are likely enriched  
273 in iron above the global mean crustal ratio (0.44; McLennan, 2001) by approximately  
274 one-third (~36%); in agreement with previous studies suggesting that the majority of iron  
275 in grassland and forest fires comes from the surrounding soils, either directly or via

276 resuspended particles previously deposited on vegetation (e.g., Gaudichet et al., 1995;  
277 Maenhaut et al., 1996; Paris et al., 2010). Compared to ordinary dust storms, fire-driven  
278 (pyro-convective) dust emission allows for (a) increasing emissions of larger particles, (b)  
279 entrainment at higher altitudes (Veira et al., 2015), and, if entrained in the free  
280 troposphere or above, (c) longer atmospheric travel distances and associated processing,  
281 e.g., 2020 Australian megafire smoke travelled across the Southern Ocean to South  
282 America (Khaykin et al., 2020). In addition to the heat generated in fires determining the  
283 strength of pyro-convective upwinds, high temperatures can alter soil properties and  
284 morphology such as mineralogy, texture, porosity, particle size distribution, and water  
285 capacity (Atanassova and Doerr, 2011; McNabb and Swanson, 1990; Pérez-Cabello et  
286 al., 2006). The coupling of fires and dust emission processes likely constitutes a large  
287 source of uncertainty for marine biogeochemical studies and a first estimate of enhanced  
288 global dust aerosol emission fluxes, both during and following biomass burning events,  
289 are discussed in Sidebars 1 and 2, respectively.

290 Nitrogen emissions from fires species are well characterized in the literature (e.g., Andrea  
291 2019) but estimates of other nutrients are missing. Overall, fire iron and phosphorus each  
292 contribute <1% of the total mass emitted in fires (Reid et al., 2005). Mahowald et al. (2005)  
293 and Luo et al. (2008) used Amazonian biomass burning observations of phosphorus and  
294 iron, respectively, to estimate emission ratios relative to black carbon (BC). As BC is a  
295 product of combustion and a core component in atmospheric models this provided a  
296 convenient methodology to estimate global phosphorus and iron emissions from fires  
297 (Mahowald et al., 2008; Myriokefalitakis et al., 2016, 2018). Here, we extend this analysis  
298 beyond the Amazonian region and find a mean Fe/BC ratio of 0.30-0.41 and P/BC ratio

299 of 0.016-0.12, suggesting that iron is being emitted at least double the rate of phosphorus  
300 in fires (Tables S3 and S4). Particle size segregated observations are more limited but  
301 suggest coarse sized emissions of phosphorus and iron are emitted at significantly higher  
302 ratios (mean across studies: 0.026-0.25 and 0.74-0.80, respectively) than fine sizes  
303 (mean across studies: 0.0013-0.010 and 0.019-0.032, respectively). Lower ratios were  
304 derived using observational data to calculate linear regression coefficients (Figures S2  
305 and S3) while higher values are the mean across reported literature values from only  
306 those studies conducted near fire activity or reporting dry (biomass burning) season  
307 values (Table S3). The lower P/BC ratios are similar to those reported by Mahowald et al.  
308 (2005). Likewise, the lower fine Fe/BC ratio is similar to Luo et al. (2008) but the lower  
309 coarse mode ratio is a little over half the value they reported of 1.4 based on less  
310 observational data than used here. As dust emissions are mostly coarse sized particles,  
311 while combustion emissions are generally fine sized, this suggests a major presence of  
312 iron and phosphate from dust in fire plumes. Forest and/or less energetic (“smouldering”)  
313 fires may also release more nutrients compared to grassland and/or intense (“flaming”)  
314 fires (Figure S4). However, observations comparing nutrient content in aerosol from  
315 flaming and smouldering fires are scarce. Additional study is necessary to understand  
316 what fire characteristics determine their potential for nutrient delivery.

317 An early estimate by Guieu et al. (2005) estimated at  $\leq 10\%$  the global marine soluble iron  
318 deposition comes from fires. Increased observations and process understanding lead to  
319 an updated estimate of around 20% (18.4–22.5%; Table S2). Regionally and during the  
320 burning (dry and warm) season, this contribution can be higher, e.g., Southern  
321 Hemisphere soluble atmospheric iron and phosphorus deposition from fires and dust may

322 be similar (Barkley et al., 2019; Hamilton et al., 2020b). A linkage between fire emission  
323 and enhanced aerosol solubility has been suggested downwind to burning regions,  
324 however this mechanism remains unclear (Mahowald et al., 2018; Paris et al., 2010;  
325 Perron et al., 2020). In the case of iron, studies have estimated solubility from fire sources  
326 to be between 2% and 46%. The lowest solubilities (2%) were observed from southern  
327 Europe and west African sources (Guieu et al., 2005; Paris et al., 2010), mid-range  
328 solubilities are observed from Australian sources at 18% (Bowie et al., 2009) or  
329 estimations based on wood composition range between 10% to 15% (Rathod et al.,  
330 2020), and the highest (46%) come from a southeastern USA controlled burn laboratory  
331 study (Oakes et al., 2012). The large range in reported fire iron solubilities has led to  
332 differing representations within models. Some investigations propose that fire iron  
333 solubility is 0% at emission with acidic and organic compounds co-emitted in fires  
334 significantly enhancing solubility of both fire and dust iron sources during transport of the  
335 smoke plume (Ito, 2015). Others have chosen to apply higher initial fractional solubility to  
336 fire iron (compared to dust sources), e.g., 33% and 4% for fine and coarse particle sized  
337 emissions, respectively (Hamilton et al., 2019). The postulate of fire plumes from  
338 Australian sources being a nutrient source to downwind oceanic regions was recently  
339 supported by field measurements of high aerosol iron solubility in concurrence with  
340 intrinsic or remotely sensed indications of the presence of aerosol from fires in samples  
341 collected at several places across Australia (Perron et al., 2020; Strzelec et al., 2020;  
342 Winton et al., 2016). When implementing such field observations into an inverse model,  
343 the minor source of iron from Australian bushfires was estimated to represent the  
344 dominant source of soluble iron (up to 82% regionally) due to higher solubility compared

345 to mineral dust source (Ito et al., 2020a). We further explore the role of Australian  
346 (wild)fires on Southern Ocean biogeochemical cycles in a case study below.

### 347 **Volcanic Aerosol Impacts on Marine Nutrient Supply**

348 Volcanic eruptions release large volumes of volcanic ash and aerosols into the  
349 atmosphere, making them an important natural nutrient source to the open ocean,  
350 especially on local-to-regional scales. The physicochemical properties and depositional  
351 patterns of volcanic aerosols differ from mineral dust, wildfires, or anthropogenic sources  
352 (Langmann, 2013). Volcanic ash is a size class of fragmented particles with diameters of  
353 submicron to  $\leq 2$  mm; such a large particle size range affects settling velocities and hence  
354 the nutrient flux to receiving water columns. The chemical composition, surface salt  
355 coatings, and particle size distribution of volcanic ashes varies widely and so does the  
356 array of nutrients (Duggen et al., 2007; Olgun et al., 2013) and trace metals (Hoffmann et  
357 al., 2012; Mahowald et al., 2018) released.

358 The volcanic ash supply to marine environments are as high as  $100 \text{ g m}^{-2}$  in the vicinity  
359 of the volcano then decreasing exponentially to values of  $0.1\text{-}0.3 \text{ g m}^{-2}$  several hundreds  
360 of kilometers away in open ocean regions (Olgun et al., 2013). The most important  
361 volcanic ash deposition regions are the Equatorial Eastern Pacific, sourced by the Central  
362 American Volcanic Arc, the Northwestern Pacific receiving the influence of the Kamchatka  
363 and the Aleutian Islands and the Southwestern Pacific both supplied by the South  
364 American volcanic arcs (Figure 5). Estimates based on marine sediment core data show  
365 that  $128 - 221 \times 10^{15} \text{ g}$  of volcanic ash has been deposited during the last millennium into  
366 the Pacific Ocean (covering  $\sim 70\%$  HNLC regions). Remote volcanic hotspots, such as

367 Iceland and Hawaii, are also important nutrient sources to the North Atlantic and the North  
368 Pacific (Figure 5).

369 In addition to volcanic ammonium-bearing clinopyroxene minerals, intense lightning  
370 within volcanic eruption plumes and atmospheric processing are both important sources  
371 of nitrogen bearing compounds on ash surfaces. The mineral composition of volcanic  
372 ashes constitutes 45-75 wt% silica and 1.0-11.0 wt% iron, depending mainly on the  
373 chemistry of the lava and eruption type. Iron solubility of volcanic ash varies significantly  
374 with the highest measured in acidic (pH 1-5) solutions at 22% and lower measured in  
375 buffered seawater at 0.001-1.8% (Duggen et al., 2010). There is no positive correlation  
376 between the iron content of ashes and the amount of iron released into the seawater (57-  
377 314 nmol Fe release per gram of ash; Olgun et al., 2011), and the most likely source for  
378 the rapid iron release is suggested to be the soluble surface salt coatings in the form of  
379 iron-bearing halides that are formed within the eruption plume (Duggen et al., 2010).

### 380 **Anthropocene Perturbations to Aerosol Nutrient Sources**

381 Increased industrial, transport, mining, and agricultural activity provide new nutrient  
382 emission sources to the atmosphere, imposing a significant increasing trend in nutrient  
383 (and toxicant) deposition to marine ecosystems (Hamilton et al., 2020b; Ito, 2015; Jickells  
384 et al., 2017; Luo et al., 2008; Matsui et al., 2018; Myriokefalitakis et al., 2020). Additionally,  
385 natural emissions are impacted by human activity through anthropogenic land use and  
386 climate changes. For example, since the Industrial Revolution dust emissions may have  
387 doubled (Mahowald et al., 2010) while fire emissions may have halved (Hamilton et al.,  
388 2018). Overall, the hemispheric balance of aerosol nutrient deposition, and thus of NPP,

389 has likely shifted positively towards the Northern Hemisphere since the Industrial  
390 Revolution (Hamilton et al., 2020a; Jickells et al., 2017; Myriokefalitakis et al., 2020).

391 Anthropogenic activity has introduced two new phosphorus sources: industrial  
392 combustion and fertilizer production and use. Global combustion sources are estimated  
393 to release ~22 Gg phosphate annually assuming 50% solubility at emission (Mahowald  
394 et al., 2008; Myriokefalitakis et al., 2016). Fertilizer production-related emissions occur  
395 during phosphate rock processing, drying, and storage, and can be mostly considered  
396 bioavailable due to their intended end use for plant uptake in agricultural fields. This latter  
397 new source is currently not included in phosphorus biogeochemistry studies, but we  
398 suggest it could be considerably higher (on average 530 Gg a<sup>-1</sup>; see SI Methods for  
399 details) than combustion sources and double the amount of all other phosphate source  
400 emissions (273 Gg a<sup>-1</sup>) combined (Mahowald et al., 2008; Myriokefalitakis et al., 2016).  
401 The relative source contributions of phosphate thus appear very different when  
402 considering this source (Figure 3b).

403 Apportioning the fraction of aerosol nutrient burdens within the atmosphere to  
404 anthropogenic sources has been tentatively quantified in several recent studies (e.g.,  
405 Jickells et al., 2017; Lamb et al., 2021; Matsui et al., 2018). In particular, the use of  
406 nitrogen and iron isotope fractionation differences could aid in distinguishing a human  
407 versus natural source signal in situ (Altieri et al., 2021; Conway et al., 2019), although  
408 additional fractionation associated with atmospheric chemistry during transport can also  
409 influence measured values. While anthropogenic aerosol is, in general, pervasive  
410 throughout the globe, some pristine aerosol regions may exist and are most likely to occur  
411 in summertime over HNLC marine regions (Hamilton et al., 2014; Uetake et al., 2020).

412 That HNLC marine regions are co-located with those atmospheric regions least impacted  
413 by human activity in the present suggests that future anthropogenic activity hotspots could  
414 impact biogeochemical cycles if developed upwind of pristine regions (Hamilton et al.,  
415 2020a; Myriokefalitakis et al., 2020). Future NPP is predicted to reduce owing to  
416 increased ocean stratification under future warming ocean conditions (Wang et al., 2015).  
417 Increased anthropogenic nutrient fluxes may (partially) offset such reductions, however,  
418 if human activity preferentially increases nitrogen and phosphate deposition, marine  
419 ecosystems may further shift towards iron or other micronutrient limitation (Letelier et al.,  
420 2019; Mahowald et al., 2018).

#### 421 **The Organic Nutrient Fraction**

422 Observations compiled by Kanakidou et al. (2012) suggest that ~35% of both aerosol  
423 nitrogen (3-90%) and phosphorus (20-83%) could be organic in nature, while modeling  
424 suggests a corresponding deposition flux fraction of around ~20-25% for nitrogen  
425 (Kanakidou et al., 2016) and up to ~50% for phosphorus (Myriokefalitakis et al., 2016).  
426 Recent biogeochemistry model calculations showed that when organic nutrients are  
427 considered, substantial increases in nitrogen-fixation was simulated in the tropical Pacific  
428 and Atlantic Oceans but it was balanced by decreases elsewhere (up to ~40%) due to  
429 the additional nitrogen inputs through organics (Myriokefalitakis et al., 2020). Although  
430 the overall impact of atmospheric organic nutrient inputs to the ocean on marine NPP is  
431 generally estimated low on a global scale (~2.4%), stronger regional changes are  
432 calculated within the oligotrophic subtropical gyres, where the additional atmospheric  
433 nitrogen deposition can support extra production of up to 15-20% (Myriokefalitakis et al.,  
434 2020).

435 The atmospheric organic nitrogen fraction has a strong anthropogenic component while  
436 organic phosphorus is mainly found in natural phosphorus-bearing aerosols such as  
437 bioaerosols (Figure 3a). Primary biogenic particles are leaf pieces, bacteria, fungi spores,  
438 or pollen released into the atmosphere either deliberately by biota or accidentally through  
439 entrainment by strong winds (Després et al., 2012; Jaenicke et al., 2007; Mahowald et  
440 al., 2008). Although poorly studied, most authors suggest higher concentrations above  
441 high productivity forests. For example, above the Amazon forest, estimates suggest 30%  
442 of the  $<10\mu\text{m}$  particles are primary biogenic particles (Graham et al., 2003). Since most  
443 life forms have elevated phosphorus (Redfield ratios suggest  $\sim 0.5\%$ ) compared to crustal  
444 content ( $<0.1\%$ ), primary biogenic particles are thought to be an important source of  
445 phosphorus to the atmosphere, although much of this phosphorus is enclosed within large  
446 particles which fall close to the emission source (Brahney et al., 2015; Tipping et al.,  
447 2014). Biogenic particles are likely to contain mostly bioavailable phosphorus, and thus  
448 they may represent the most important source of soluble phosphorus to many ocean  
449 regions (Myriokefalitakis et al., 2016).

450 On the other hand, organic-bound iron is produced during transport when iron-containing  
451 aerosols undergo organic ligand-mediated dissolution processes (i.e., as Fe(II/III)-oxalate  
452 complexes). Of atmospheric organic ligands, oxalic acid is currently considered the most  
453 important species and is used as a proxy for organic ligand iron-dissolution processes  
454 since: (1) it is thought to be the most abundant in the atmosphere and (2) it is the most  
455 effective ligand in promoting iron solubilization through the formation of iron–oxalate  
456 complexes at the mineral's surface which polarize and weaken Fe–O bonds (e.g., Ramos  
457 et al., 2014). Oxalic acid is produced within the atmosphere by aqueous-phase

458 photochemical processes, mainly in cloud droplets (e.g., Myriokefalitakis et al., 2011).  
459 Biogenic VOCs are the most important precursors and owing to a strong source, intense  
460 photochemistry, and a strong convective transport potential, high oxalate (the  
461 deprotonated form of oxalic acid) concentrations are found in tropical regions  
462 (Myriokefalitakis et al., 2011). However, dicarboxylic acids in the atmosphere may also  
463 have various primary sources including biomass burning, vehicular exhausts, and cooking  
464 emissions. Dicarboxylic acids show, nevertheless, a strong correlation with elemental  
465 carbon and levoglucosan (Cao et al., 2017; Cong et al., 2015) and high concentrations of  
466 oxalic acid have been observed during the Amazonian burning season (Kundu et al.,  
467 2010). This suggests that biomass burning (rather than coal combustion or vehicular  
468 exhaust) could be an additional atmospheric source of oxalate (Schmidl et al., 2008;  
469 Yamasoe et al., 2000). Some studies postulated that high temperatures generated by  
470 fires may catalyze the transformation of insoluble iron oxides in soils into more labile forms  
471 in the presence of organic matter (Ito et al., 2018). Such modified soils are entrained into  
472 the atmosphere both during and post-fire (Sidebars 1 and 2).

### 473 **Australia: A Major Nutrient Source to Southern Hemisphere Oceans**

474 Australia is one of the largest arid regions in the Southern Hemisphere, along with  
475 Patagonia in South America and the Kalahari and Namib deserts in southern Africa. The  
476 central 'Outback' region of Australia is composed of large sandy deserts towards the west  
477 and arid geological basins (lake Eye and Murray-Darling) on the east. Australian  
478 atmospheric circulation is divided into three main aerial pathways, potentially irrigating:  
479 (1) the Indian Ocean via the "north-west dust path"; (2) the Tasman Sea and southern  
480 Pacific downwind from the "south-east dust path"; and (3) the Great Australian Bight

481 through a smaller atmospheric depression blowing south-westwards across Western  
482 Australia southern coast. Large dust storms frequently emphasize the uplift and transport  
483 processes of iron-bearing dust in Australia, with mega-tons of red soil particles being  
484 carried away to surrounding marine areas (Gabric et al., 2010; Mackie et al., 2008).  
485 Current modeling projections suggest that Australia accounts for 10% (4-30%) of dust  
486 deposited into the Southern Ocean, 7% (3-11%) of dust deposited into the Indian Ocean,  
487 and 68% (20-81%) of dust deposited into the South Pacific (Kok et al., 2021).

488 In addition to being a major dust source, southern Australia is frequently hit by devastating  
489 summertime wildfires, e.g., the summers of 2002-03, 2005-06, 2006-07, 2009 and 2012-  
490 13 and 2018-19 (Tasmania only) (Bureau of Meteorology, <http://www.bom.gov.au/>).  
491 Recently, the 2019-20 Australian megafires had an unprecedented impact on Australia's  
492 vegetation, burning no less than 21% of the country's temperate and broadleaf forests  
493 (Boer et al., 2020). Fire-induced pyrocumulonimbus clouds associated with this natural  
494 disaster have been followed across the whole Southern Hemisphere, evidencing major  
495 and long-lasting (over months) perturbations of the atmospheric composition (including  
496 the stratosphere) in the entire Southern Hemisphere. The magnitude and elevation  
497 attained by the 2020 Australian fire plume was compared to atmospheric perturbation  
498 following the strongest volcanic eruption of the past 30 years (Khaykin et al., 2020).

499 Recent time series (2016-2020) observations, from samples collected at Mt. Wellington  
500 in Tasmania (42.9°S, 147.2°E), reveals the aeolian transport of nutrients from Australia  
501 southwards and into the Southern Ocean. Total and soluble iron measurements are  
502 compared with two atmospheric iron models in Figure 6a: IMPACT (Ito et al., 2021) and  
503 MIMI (Hamilton et al., 2020b) (SI Methods); see SI Methods. The four-year field

504 observations show that, during burning periods, atmospheric iron loading is significantly  
505 higher than during non-fire days regardless of the season. This further suggests that  
506 wildfires are a non-negligible source of iron to Australian atmospheric transport pathways.  
507 While biomass contains small amounts of iron, this raises the question of which  
508 mechanism leads fires emission to entrain a significant number of iron-bearing particles  
509 within their plumes (Figures 4 and 5, and Sidebars 1 and 2).

510 Dust alone requires extensive atmospheric processing to reach high aerosol iron solubility  
511 measured in field observations - often exceeding 10% over much of the Southern  
512 Hemisphere (Baker et al., 2013; Perron et al., 2020; Winton et al., 2015). Iron modeling  
513 currently includes dust, anthropogenic combustion, and fire sources of iron  
514 (Myriokefalitakis et al., 2018). Comparison of ensemble modeling to Mt. Wellington  
515 observations suggest fidelity in representing the total mass of iron emitted from Australia  
516 (Figure 6a). However, while the observed soluble iron range on fire-days (defined when  
517 intrinsic measurement of levoglucosan  $>10 \text{ ng m}^{-3}$ ) is captured, the median ( $2.5 \text{ ng m}^{-3}$ )  
518 is lower than that observed ( $4.0 \text{ ng m}^{-3}$ ). Furthermore, modeling significantly  
519 underestimates median iron solubility for non-fire days (observed: 5.9%, IMPACT: 1.2%,  
520 MIMI: 3.1-3.6%), suggesting (1) the existence of a currently unaccounted for, and  
521 important, atmospheric source of highly soluble iron emitted on these days or (2) that  
522 processing of dust iron happens at a faster rate than currently realized under pristine  
523 conditions.

524 The contribution of each source to soluble iron concentrations between models is different  
525 (Figure 6b); IMPACT emphasizes dust while MIMI emphasizes fires. The increased  
526 contribution of fires to soluble iron concentrations on fire-days is also higher in MIMI

527 compared to IMPACT (51-79% increase versus 38%). As near future predictions warn for  
528 an increasing occurrence of conditions prone to the ignition of megafires in temperate  
529 regions of Australia (Dowdy et al., 2019; Di Virgilio et al., 2019), more constraint on the  
530 properties of fire sourced iron, including its pyrogenic fraction, is therefore needed.

531 We quantify the impact of the 2020 Australian megafires on ocean biogeochemistry using  
532 the PISCES model (Aumont et al., 2015) and compare with previous years (Figure 6c).  
533 To isolate fire impacts the model was run twice (SI Methods), with and without fire iron  
534 deposition, and the difference between them analyzed for January-February (Australian  
535 peak fire activity). Model soluble iron from fire was predicted using daily QFED emissions  
536 as this represented the maximum Australian emission estimate and also simulated  
537 median soluble iron concentrations closest to observations (observations:  $4.0 \text{ ng m}^{-3}$ ,  
538 IMPACT:  $0.9 \text{ ng m}^{-3}$ , MIMI-FINN:  $2.4 \text{ ng m}^{-3}$ , MIMI-QFED:  $4.5 \text{ ng m}^{-3}$ ). Large variability in  
539 fire activity between years increases Southern Ocean ( $30\text{-}65^\circ\text{S}$ ) NPP between 1.1% and  
540 3.5%, with the second largest increase, by 3.0%, occurring in 2020. As fires also  
541 represent a large source of nitrogen and phosphate (Figure 3a), 3% additional NPP may  
542 be a conservative estimate of the potential marine stimulation following atmospheric  
543 deposition of fire aerosol. In earlier modeling work, the combined addition of nitrogen and  
544 iron together (such as in fire plumes) was shown globally to induce larger NPP than a  
545 single element supply would (Krishnamurthy et al., 2009).

546 In these experiments fire deposition of soluble iron has a clear overall positive impact on  
547 NPP in subarctic and Southern Ocean HNLC regions (Figure 7). Depending on year, fires  
548 increased annual global NPP up to 0.7% (Southern Ocean: 0.7-1.3%, central Pacific  
549 ( $30^\circ\text{S}$ - $30^\circ\text{N}$ ): -0.3-1.2%, north Pacific ( $45\text{-}65^\circ\text{N}$ ): 0.1-1.3%). In some regions, such as the

550 equatorial Pacific, NPP increases are balanced by decreases downstream, a result also  
551 seen in other studies and suggested to be linked to macronutrient decreases downstream  
552 from where iron fertilization occurs (Hamilton et al., 2020a; Ito et al., 2020b; Tagliabue et  
553 al., 2008). Improved understanding of how variability in aerosol nutrient supply, over  
554 different timescales, impacts ocean biogeochemistry requires a holistic multidisciplinary  
555 approach, including consideration of: the physicochemical properties of different nutrient  
556 aerosol sources; those interactions during atmospheric transport altering bioavailability;  
557 mixing of aerosol nutrients from varied sources; and the multiple processes operating in  
558 the surface mixed layer post-deposition.

## 559 **Summary Points**

- 560 1. Major nutrient sources include deserts/soils, wildfires, volcanoes, biogenic  
561 particles, and industrial or vehicular pollution. Sources of nitrogen are strongly  
562 linked to human activity while phosphate and soluble iron have considerable  
563 natural sources.
- 564 2. Northern Hemisphere ocean nutrient fluxes are dominated by mineral dust sources  
565 with an additional anthropogenic source at (sub-)polar latitudes. In the equatorial  
566 Pacific and Southern Hemisphere oceans, fires are likely to be important.
- 567 3. Two-thirds of the iron in fires could be associated with dust particles entrained  
568 within smoke plumes, although more work is needed to quantify this contribution.  
569 Additionally, fires transform the landscape (soils and vegetation), creating a legacy  
570 of secondary impacts altering nutrient aerosol supply until the ecosystem recovers.
- 571 4. In situ aerosol addition experiments suggest atmospheric deposition favors an  
572 increase (~+80%) in Chl-a. However, in many experiments no response was

573 recorded and occasionally aerosols act as a pollutant or ballast decreasing Chl-a.  
574 The characteristics of the receiving body of water, including its biota, is thus crucial  
575 to understanding the impacts of changes to aerosol nutrient delivery.

- 576 5. At the global annual mean, modeling suggests fires have a net positive impact on  
577 NPP (+0.7%). Regional increases can be higher (e.g., Southern Ocean: 0.7-1.3%)  
578 and peak during the burning season (e.g., January-February Australian megafires  
579 could have increased Southern Ocean NPP up to +3%).
- 580 6. A significant non-fire source of Australian soluble iron or mechanism increasing  
581 the solubility of dust may be missing from atmospheric simulations of the iron cycle.

## 582 **Future Issues**

- 583 1. What are the mechanisms by which aerosol nutrients are distributed to oceans  
584 under different climate regimes?
- 585 2. How does regional land use and climate change alter dust and (wild)fire activity,  
586 both historically and in the future, and their roles in global biogeochemical cycles?
- 587 3. What field observations can aid in linking aerosol sources to observed  
588 characteristics? Some suggestions include: Isotopes show promise in quantifying  
589 human versus natural source contributions within mixed airmasses, but targeted  
590 sampling on emissions from fires, forests, volcanoes, urban environments,  
591 deserts, and mining operations are all essential; Time series stations can quantify  
592 natural variability and the impact of long-term anthropogenic surface disturbances  
593 as well as aid in extreme event attribution studies; Aircraft measurements taken  
594 over marine regions compliment current ship-based observations to build a profile  
595 of the distribution and changing properties of aerosol nutrients over their

596 atmospheric lifetime; In situ bioassay experiments are highly valuable, particularly  
597 in unexplored key areas of the ocean.

598 4. How do differences in the physicochemical nature of different aerosol types alter  
599 their interactions with a similarly diverse suite of oceanic physical, chemical, and  
600 biological processes?

601 5. What is the response of net primary production and the biological pump to changes  
602 in aerosol fluxes from different sources in the context of a changing ocean physical  
603 state?

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#### 1189 **SIDEBAR#1. The interface between fires and dust: in-plume dust entrainment**

1190 The impact of different agricultural-related fires on near-surface wind patterns was  
1191 investigated by the idealized model study of Wagner et al. (2018), which found  
1192 aerodynamic conditions suitable for dust emission in the vicinity of such fires. The  
1193 simulated fire-modulated winds were coupled with two different dust emission  
1194 parameterizations to obtain first estimates of the effectiveness of pyro-convectively driven  
1195 dust emissions (Wagner et al., submitted to JGR). Depending on the model setup, fire-  
1196 driven dust emission fluxes of 1.0-5.0 g m<sup>-2</sup> h<sup>-1</sup> were found, which would, if scaled up  
1197 globally, account for up to 20 Tg a<sup>-1</sup>. So, with respect to the total global dust emissions,  
1198 estimated between 700 and 3600 Tg a<sup>-1</sup> in models with comparable particle size (0.06–  
1199 20 μm) (Wu et al., 2020), pyro-convectively driven dust emission could contribute an  
1200 additional 3-14%, especially in regions outside the “dust-belt”. Assuming a dust iron  
1201 fraction of 3.5% and that 64% fire-iron is dust sourced gives emissions (<20 μm particle  
1202 size) of 1.1 Tg Fe a<sup>-1</sup> from fires, which is the same as a previous estimate by Luo et al.  
1203 (2008).

#### 1204 **SIDEBAR#2. The interface between fires and dust: post-burn landscape**

1205 The strength and duration of post-fire dust emissions can vary significantly depending on  
1206 the geographical location, the meteorology, and the re-vegetation period. Results from

1207 several studies reflecting the geographical diversity point towards a large variability of  
1208 enhanced dust fluxes between individual sites (Dukes et al., 2018; Jeanneau et al., 2019;  
1209 Whicker et al., 2006). By using their averaged results as a proxy for post-burnt landscapes  
1210 and assuming the periods without vegetation permitting wind erosion (months-to-year), it  
1211 might be possible that such post-fire emissions can contribute in the order of magnitude  
1212 of 100 Tg a<sup>-1</sup> of soil dust emissions, with a possible uncertainty interval of the same order  
1213 of magnitude. However, post-fire sources might at least be partly considered for global  
1214 estimates on dust emission fluxes already

1215

#### 1216 **Terms and Definitions List:**

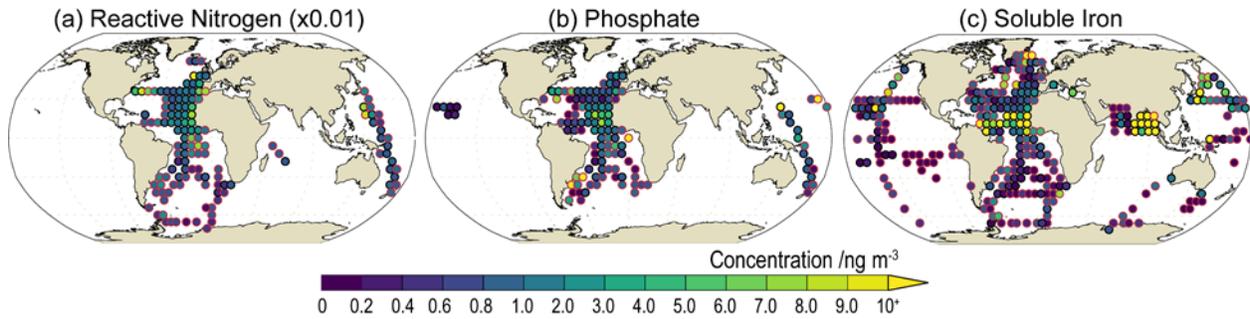
- 1217 • Aerosol: solid or liquid particles suspended in the atmosphere.
- 1218 • Biogeochemical cycles describe how nutrients move through both the biotic and  
1219 abiotic parts of the Earth System.
- 1220 • Major anthropogenic emissions are produced by vehicular transportation, fossil  
1221 fuel and biofuel combustion, mining, intensifying agricultural practices, and metal  
1222 smelting.
- 1223 • Macronutrients: nitrogen, phosphorus, and silicate. Micronutrients: iron, zinc,  
1224 copper, cobalt, and manganese.
- 1225 • Net primary productivity (NPP): biomass accumulation rate from transformation  
1226 of energy, in excess of respiration, by photo- or chemosynthesis.
- 1227 • Surface Ocean Lower Atmosphere Study (SOLAS): <https://www.solas-int.org/>.
- 1228 GEOTRACES: <https://www.geotraces.org/>.
- 1229 • Nutrient solubility: The percentage of the total aerosol that is soluble.

- 1230 • Reactive nitrogen: oxidized, reduced, and organic forms of nitrogen which are  
1231 bioaccessible. Examples are nitrate ( $\text{NO}_3^-$ ) and ammonium ( $\text{NH}_4^+$ ).
- 1232 • Phosphorus: atmospheric transport occurs via aerosol as no stable gas phase  
1233 exists. Phosphate ( $\text{PO}_4^{3-}$ ) considered the soluble (bioaccessible) phosphorus  
1234 fraction.
- 1235 • Dust belt: latitudinal band including the primary dust source regions of North  
1236 Africa, Arabian Peninsula, and Central Asia.
- 1237 • High-Nutrient Low-Chlorophyll (HNLC) region: nitrogen and phosphorus in  
1238 excess compared to biological requirements, although atypically low chlorophyll  
1239 levels are observed.
- 1240 • Oligotrophic: depleted in nutrients and exhibiting low surface chlorophyll. Here:  
1241 Chl-a  $<0.5 \mu\text{g L}^{-1}$  excluding very coastal or upwelling studies.
- 1242 • Coarse: Particle diameter larger than 1 or 2  $\mu\text{m}$ . Fine: Particle diameter less than  
1243 1 or 2  $\mu\text{m}$ .
- 1244 • Smoldering: often persistent, low-temperature, flameless burning producing  
1245 significant smoke volumes. Flaming: higher temperature burning with flames  
1246 producing less smoke.
- 1247 • Isotopic fractionation: physical, chemical, and biological processes enrich one  
1248 isotope relative to another in predictable ways, creating distinct source  
1249 “fingerprints”.

- 1250 • Pristine aerosol region: where the post-Industrial Revolution human influence on  
1251 the aerosol state likely remains minimal.
- 1252 • Levoglucosan: an indicator of wood combustion being derived from the pyrolysis  
1253 of (hemi-) cellulose molecules.

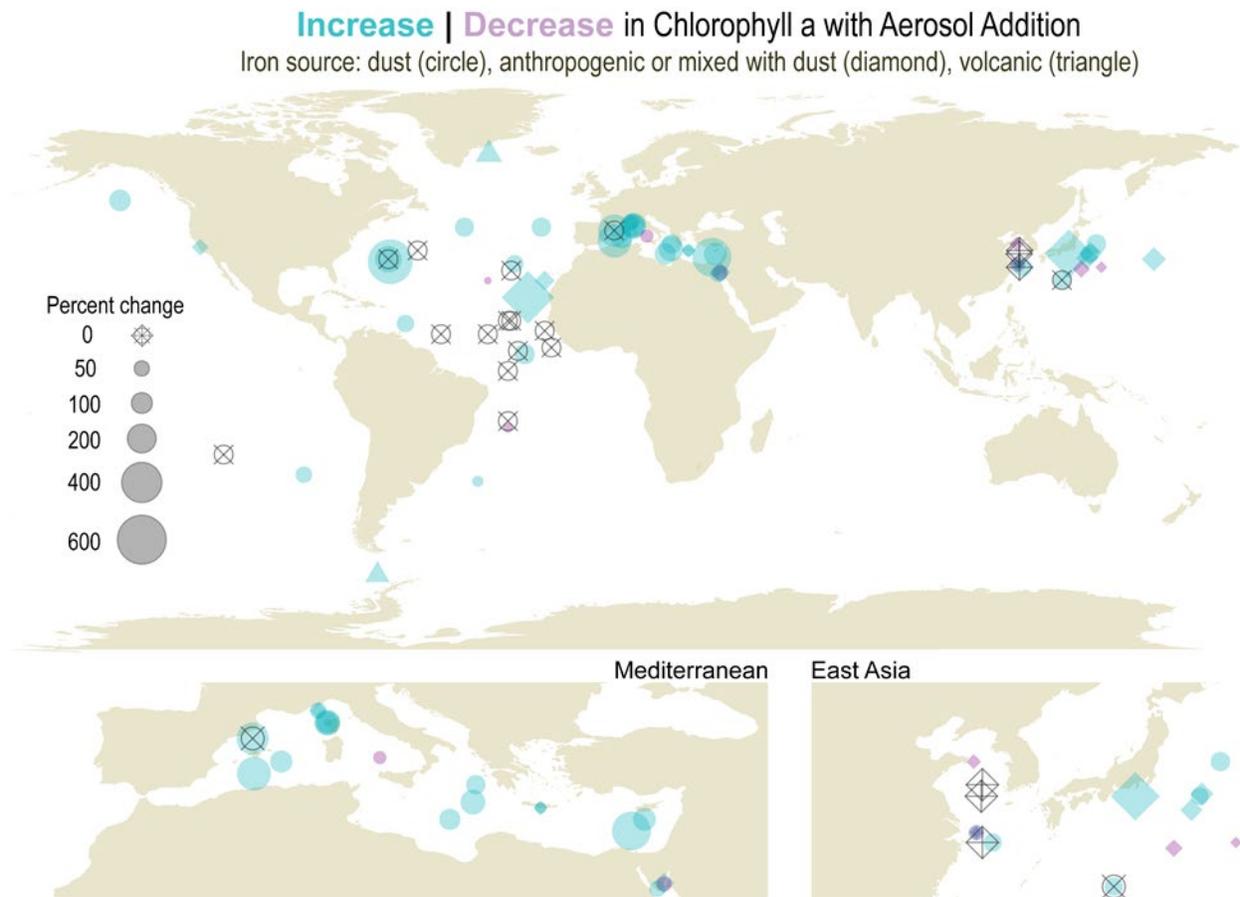
1254

1255 **Figures**



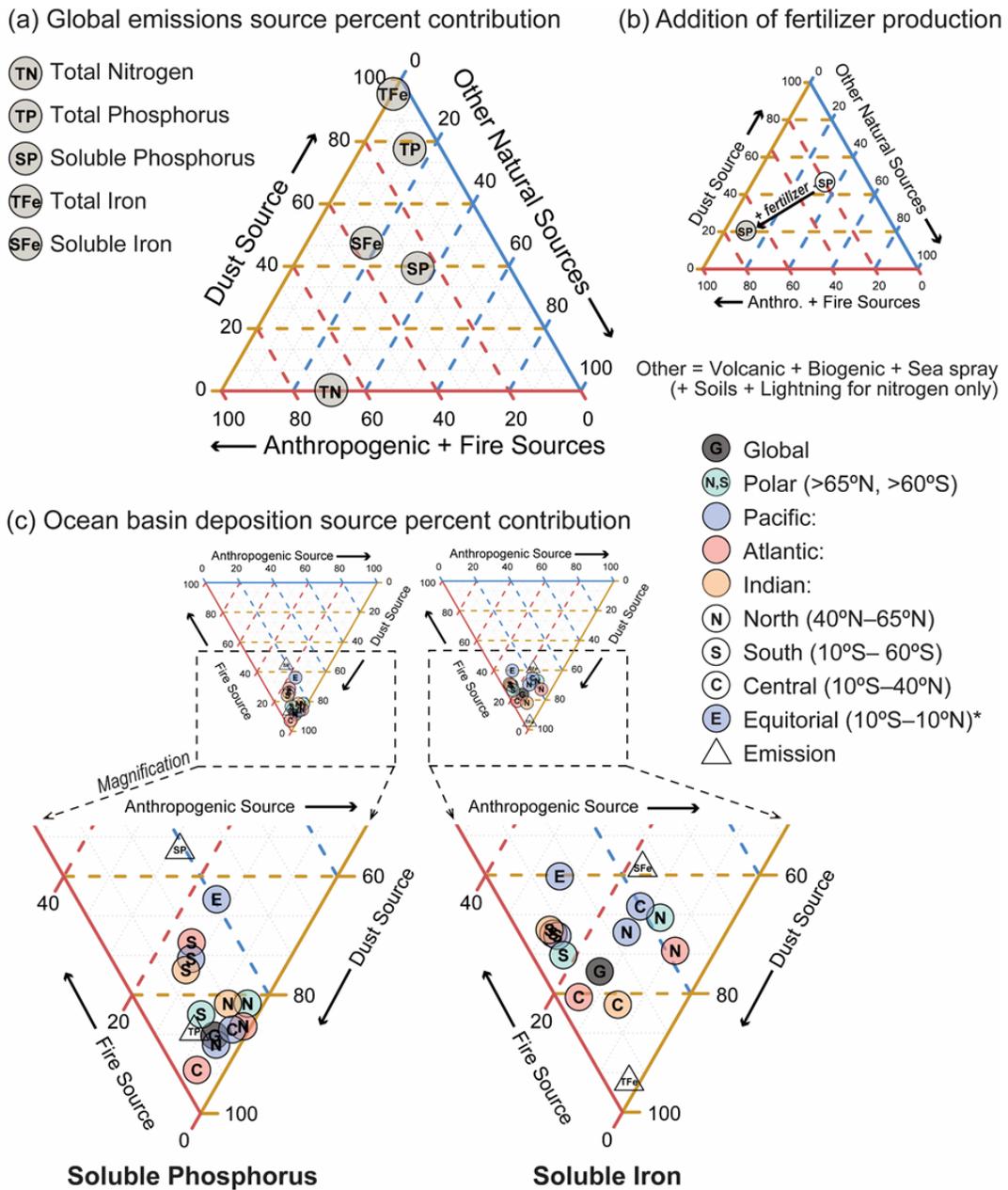
1256

1257 **Figure 1.** Ship-borne observations of atmospheric reactive nitrogen (NH<sub>4</sub> + NO<sub>3</sub> + water  
1258 soluble organic nitrogen) (a), phosphate (b), and soluble iron (c) median concentrations.  
1259 Observations aggregated over a 6x6 degree grid cell. Red outline indicates ≤2  
1260 observations, black >2 observations.



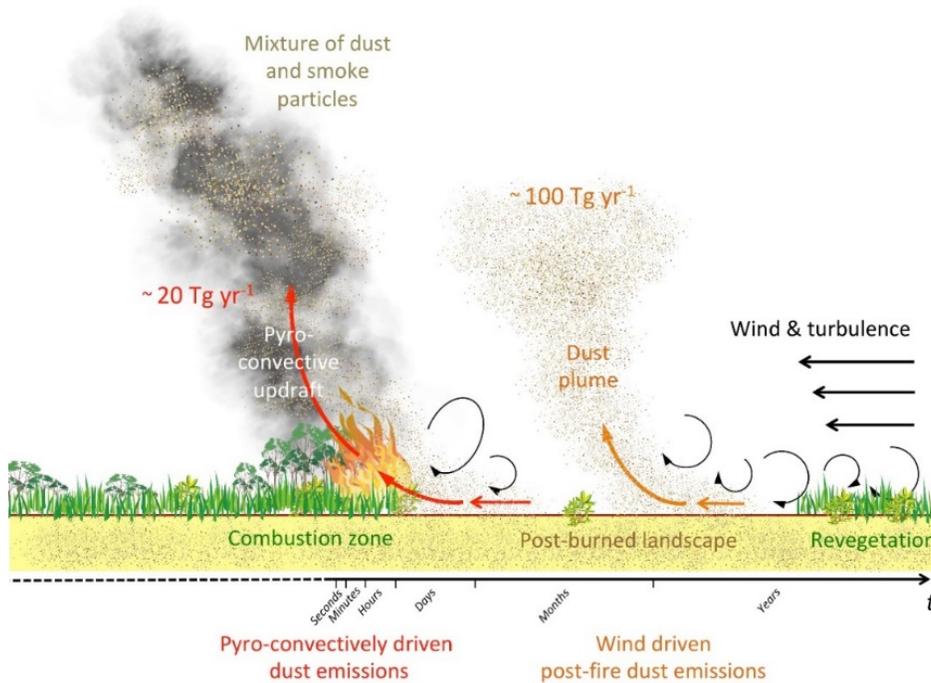
1261

1262 **Figure 2.** Maximum observed biomass change, as measured by Chl-a change relative to  
 1263 control (bearing macro- and micronutrients and metals - including potential toxic elements  
 1264 such as copper) from different sources. See Table S1 for a list of the experiments, initial  
 1265 Chl-a values, and the citations.



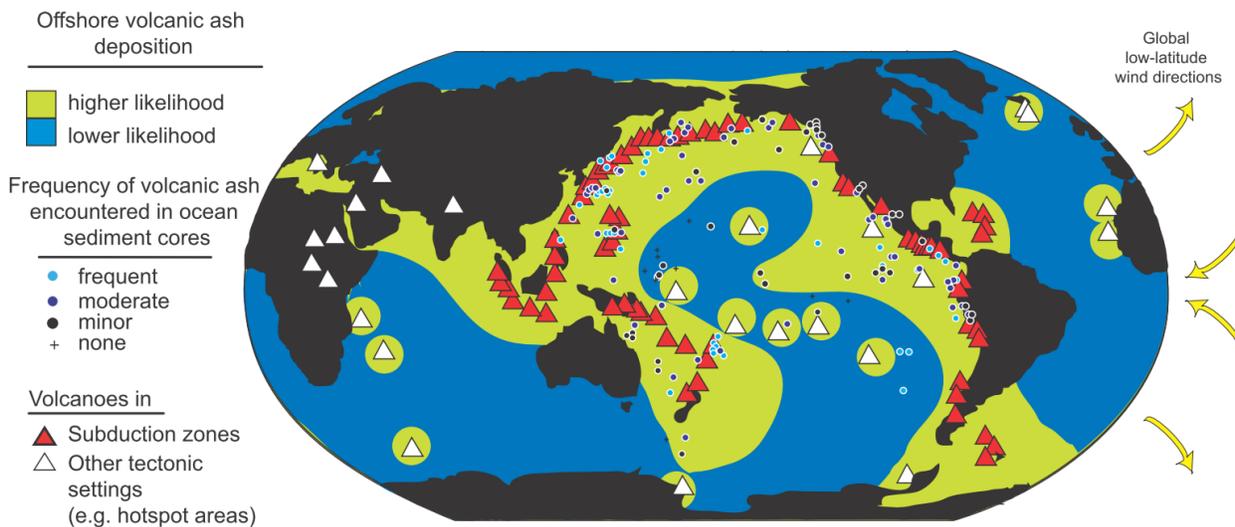
1266

1267 **Figure 3.** Percent contribution to global emissions (a), soluble phosphorus when  
 1268 accounting for fertilizer production (b), and ocean basin deposition (c) from each nutrient  
 1269 source. Emission data in (a) from Kanakidou et al., (2018) and iron modeling. Deposition  
 1270 data in (c) from modeling: phosphorus from Myriokefalitakis et al. (2020), iron from Table  
 1271 S2. (\* HNLC sub-region within the central Pacific.)



1272

1273 **Figure 4.** Properties of nutrient aerosol emissions from fire activity changes with time, as  
 1274 well as estimates in the magnitude of aerosol fluxes. See Sidebars 1 and 2 and SI  
 1275 Methods for details on global annual fluxes estimated in this study.

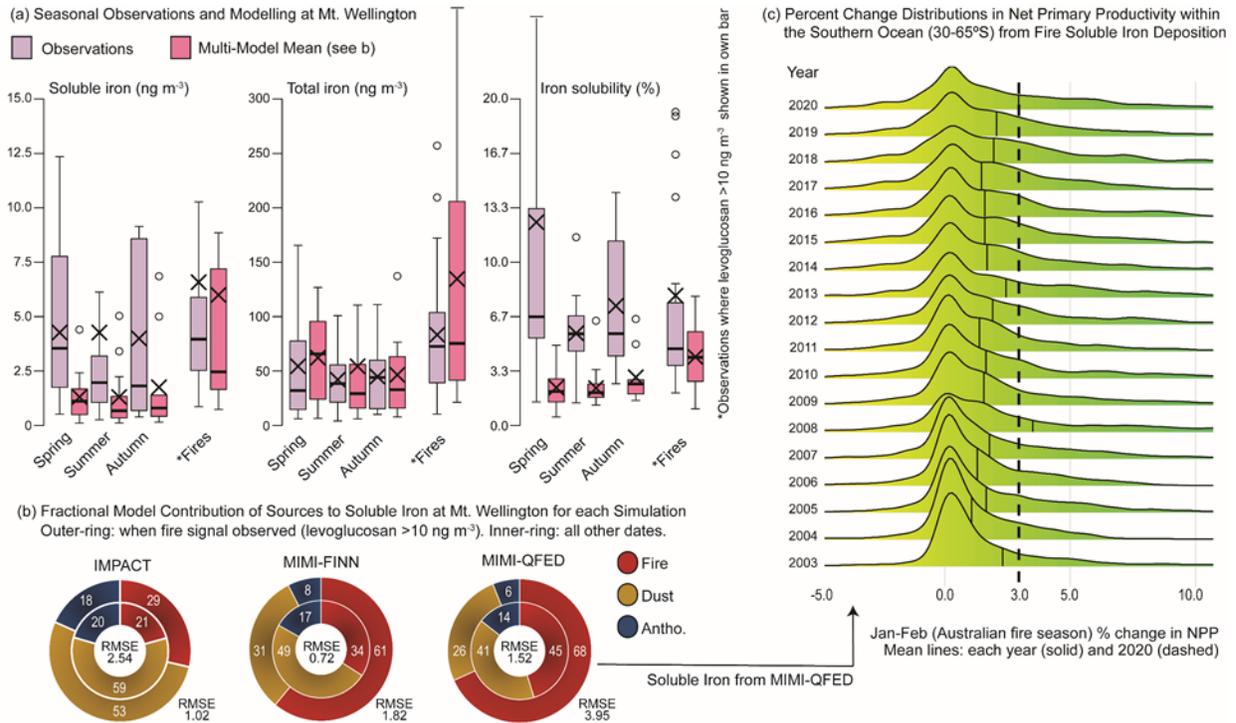


1276

1277 **Figure 5.** Global distribution of volcanoes and the offshore ash deposition areas (green  
 1278 regions) (Olgun et al., 2011). The higher likelihood of ash deposition regions shown in

1279 green color was based on the low-latitude wind directions and the frequency of volcanic  
 1280 ash layers found in the ocean drill cores (Straub and Schmincke, 1998).

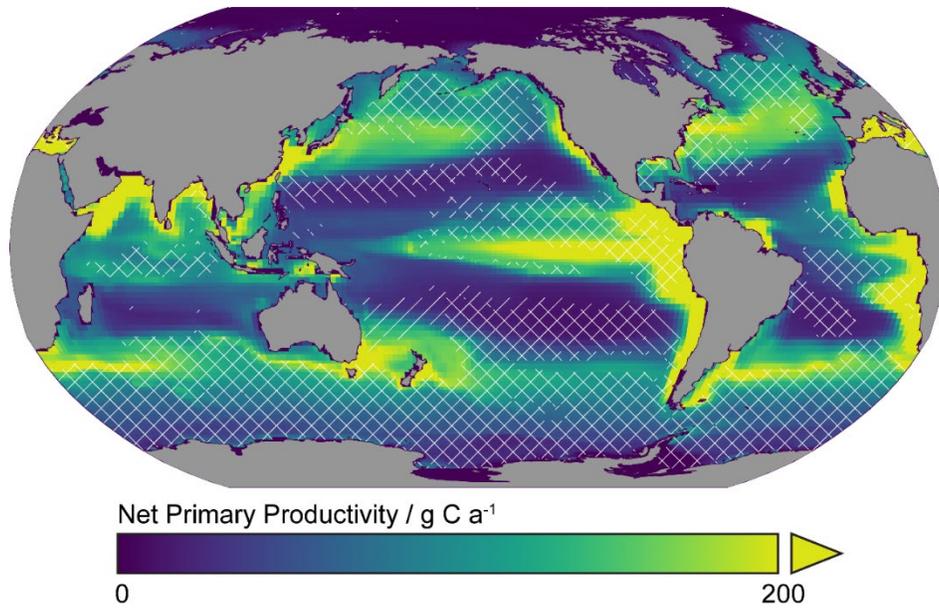
1281



1282

1283 **Figure 6.** Potential impact of wildfires on marine biogeochemistry. Observations and  
 1284 modeling of iron (a and b) and annual changes in January-February Southern Ocean (30–  
 1285 65°S) depth integrated net primary production (c). Atmospheric iron models: IMPACT (Ito  
 1286 et al., 2021) and MIMI (Hamilton et al., 2020b). The mean in (a) is shown by a cross (X).  
 1287 Levoglucosan threshold of  $10 \text{ ng m}^{-3}$  chosen to exclude residual levoglucosan levels, as  
 1288 previously reported in remote sampling regions (Bhattarai et al., 2019); therefore, fire  
 1289 observations only account for those samples displaying a distinct fire signal.

1290



1291

1292 **Figure 7.** Annual mean (2003-2019) depth integrated marine net primary productivity.

1293 White lines indicate where fire soluble iron deposition increased productivity (different fire

1294 emission datasets are indicated by orientation: 45°=QFED, -45°=FINN; no line indicates

1295 a decrease).